

## Balancing Act

Name \_\_\_\_\_

Atoms are not \_\_\_\_\_ or \_\_\_\_\_ during a chemical reaction. Scientists know that there must be the \_\_\_\_\_ number of atoms on each \_\_\_\_\_ of the \_\_\_\_\_. To balance the chemical equation, you must add \_\_\_\_\_ in front of the chemical formulas in the equation. You cannot \_\_\_\_\_ or \_\_\_\_\_ subscripts!

1) Determine number of atoms for each element.

2) Pick an element that is not equal on both sides of the equation.

3) Add a coefficient in front of the formula with that element and adjust your counts.

4) Continue adding coefficients to get the same number of atoms of each element on each side.



Mg =

Mg =

O =

O =

Try these:



Ca =

Ca =

O =

O =



N =

N =

H =

H =



Cu =

Cu =

O =

O =

C =

C =



H =

H =

O =

O =



# Balancing Act Chemical Equations Worksheet Answers

**Michal Rosen-Zvi**





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