

1.  $\underline{\quad\quad}\text{Zn} + \underline{2}\text{HCl} \rightarrow \underline{\quad\quad}\text{ZnCl}_2 + \underline{\quad\quad}\text{H}_2$
2.  $\underline{4}\text{Fe} + \underline{13}\text{O}_2 \rightarrow \underline{2}\text{Fe}_2\text{O}_3$
3.  $\underline{4}\text{P} + \underline{5}\text{O}_2 \rightarrow \underline{\quad\quad}\text{P}_4\text{O}_{10}$
4.  $\underline{\quad\quad}\text{N}_2 + \underline{3}\text{H}_2 \rightarrow \underline{2}\text{NH}_3$
5.  $\underline{\quad\quad}\text{PbO}_2 \rightarrow \underline{\quad\quad}\text{Pb} + \underline{\quad\quad}\text{O}_2$
6.  $\underline{2}\text{C}_6\text{H}_6 + \underline{15}\text{O}_2 \rightarrow \underline{12}\text{CO}_2 + \underline{6}\text{H}_2\text{O}$
7.  $\underline{\quad\quad}\text{C}_2\text{H}_5\text{OH} + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{CO}_2 + \underline{3}\text{H}_2\text{O}$
8.  $\underline{3}\text{Fe} + \underline{4}\text{H}_2\text{O} \rightarrow \underline{\quad\quad}\text{Fe}_3\text{O}_4 + \underline{4}\text{H}_2$
9.  $\underline{\quad\quad}\text{SiO}_2 + \underline{4}\text{HF} \rightarrow \underline{\quad\quad}\text{SiF}_4 + \underline{2}\text{H}_2\text{O}$
10.  $\underline{3}\text{CO} + \underline{\quad\quad}\text{FeO}_4 \rightarrow \underline{\quad\quad}\text{FeO} + \underline{3}\text{CO}_2$
11.  $\underline{\quad\quad}\text{NH}_4\text{NO}_2 \rightarrow \underline{\quad\quad}\text{N}_2 + \underline{2}\text{H}_2\text{O}$
12.  $\underline{\quad\quad}\text{N}_2\text{H}_4 + \underline{\quad\quad}\text{O}_2 \rightarrow \underline{\quad\quad}\text{N}_2 + \underline{2}\text{H}_2\text{O}$
13.  $\underline{3}\text{NO}_2 + \underline{\quad\quad}\text{H}_2\text{O} \rightarrow \underline{2}\text{HNO}_3 + \underline{\quad\quad}\text{NO}$
14.  $\underline{2}\text{KClO}_3 \rightarrow \underline{2}\text{KCl} + \underline{3}\text{O}_2$
15.  $\underline{\quad\quad}\text{C}_{12}\text{H}_{22}\text{O}_{11} + \underline{12}\text{O}_2 \rightarrow \underline{12}\text{CO}_2 + \underline{11}\text{H}_2\text{O}$
16.  $\underline{4}\text{NH}_3 + \underline{3}\text{O}_2 \rightarrow \underline{6}\text{H}_2\text{O} + \underline{2}\text{N}_2$
17.  $\underline{\quad\quad}\text{CH}_3\text{COOH} + \underline{\quad\quad}\text{O}_2 \rightarrow \underline{2}\text{CO}_2 + \underline{2}\text{H}_2\text{O}$
18.  $\underline{2}\text{C}_2\text{H}_6 + \underline{7}\text{O}_2 \rightarrow \underline{4}\text{CO}_2 + \underline{6}\text{H}_2\text{O}$
19.  $\underline{\quad\quad}\text{C}_6\text{H}_8 + \underline{10}\text{O}_2 \rightarrow \underline{8}\text{CO}_2 + \underline{4}\text{H}_2\text{O}$
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21.  $\underline{\quad\quad}\text{N}_2 + \underline{3}\text{H}_2 \rightarrow \underline{2}\text{NH}_3$
22.  $\underline{3}\text{Fe} + \underline{4}\text{H}_2\text{O} \rightarrow \underline{\quad\quad}\text{Fe}_3\text{O}_4 + \underline{4}\text{H}_2$
23.  $\underline{3}\text{Fe} + \underline{2}\text{O}_2 \rightarrow \underline{\quad\quad}\text{Fe}_3\text{O}_4$
24.  $\underline{3}\text{NH}_4\text{OH} + \underline{\quad\quad}\text{AlCl}_3 \rightarrow \underline{\quad\quad}\text{Al(OH)}_3 + \underline{3}\text{NH}_4\text{Cl}$
25.  $\underline{\quad\quad}\text{Na}_2\text{CO}_3 + \underline{\quad\quad}\text{H}_2\text{O} + \underline{\quad\quad}\text{CO}_2 \rightarrow \underline{2}\text{NaHCO}_3$
26.  $\underline{\quad\quad}\text{MnO}_2 + \underline{4}\text{HCl} \rightarrow \underline{\quad\quad}\text{MnCl}_2 + \underline{2}\text{H}_2\text{O} + \underline{\quad\quad}\text{Cl}_2$
27.  $\underline{\quad\quad}\text{PBr}_3 + \underline{3}\text{H}_2\text{O} \rightarrow \underline{\quad\quad}\text{H}_3\text{PO}_3 + \underline{3}\text{HBr}$
28.  $\underline{2}\text{ZnS} + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{ZnO} + \underline{\quad\quad}\text{SO}_2$
29.  $\underline{2}\text{NH}_3 + \underline{3}\text{O}_2 \rightarrow \underline{\quad\quad}\text{N}_2 + \underline{6}\text{H}_2\text{O}$
30.  $\underline{\quad\quad}\text{Na}_2\text{B}_4\text{O}_7 + \underline{\quad\quad}\text{H}_2\text{SO}_4 + \underline{5}\text{H}_2\text{O} \rightarrow \underline{\quad\quad}\text{Na}_2\text{SO}_4 + \underline{4}\text{H}_3\text{BO}_3$
31.  $\underline{2}\text{NH}_3 + \underline{2}\text{O}_2 \rightarrow \underline{\quad\quad}\text{NO} + \underline{3}\text{H}_2\text{O}$
32.  $\underline{4}\text{HNO}_3 \rightarrow \underline{2}\text{H}_2\text{O} + \underline{4}\text{NO}_2 + \underline{\quad\quad}\text{O}_2$
33.  $\underline{4}\text{HCl} + \underline{2}\text{HNO}_3 \rightarrow \underline{2}\text{Cl}_2 + \underline{2}\text{NOCl} + \underline{3}\text{H}_2\text{O}$
34.  $\underline{2}\text{Ca}_3(\text{PO}_4)_2 + \underline{6}\text{SiO}_2 \rightarrow \underline{6}\text{CaSiO}_3 + \underline{\quad\quad}\text{P}_4\text{O}_{10}$
35.  $\underline{\quad\quad}\text{C}_6\text{H}_{12}\text{O}_6 \rightarrow \underline{2}\text{C}_2\text{H}_5\text{OH} + \underline{2}\text{CO}_2$
36.  $\underline{2}\text{C}_8\text{H}_{18} + \underline{25}\text{O}_2 \rightarrow \underline{16}\text{CO}_2 + \underline{18}\text{H}_2\text{O}$
37.  $\underline{2}\text{CH}_4 + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{CO} + \underline{4}\text{H}_2\text{O}$
38.  $\underline{2}\text{C}_2\text{H}_2 + \underline{3}\text{O}_2 \rightarrow \underline{4}\text{CO}_2 + \underline{2}\text{H}_2\text{O}$
39.  $\underline{6}\text{H}_2\text{O} + \underline{6}\text{CO}_2 \rightarrow \underline{\quad\quad}\text{C}_6\text{H}_{12}\text{O}_6 + \underline{6}\text{O}_2$
40.  $\underline{\quad\quad}\text{Cu} + \underline{4}\text{HNO}_3 \rightarrow \underline{\quad\quad}\text{Cu(NO}_3)_2 + \underline{2}\text{NO} + \underline{2}\text{H}_2\text{O}$

# Balancing Equations Ch 7 Worksheet Answers Chapter 7

**Crispin Collins**



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