

## 12

## STOICHIOMETRY

## Practice Problems

In your notebook, solve the following problems.

## SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups (top and bottom), and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate,  $\text{KClO}_3(\text{s})$ . Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from  $3.8 \times 10^{23}$  molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

## SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(II) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 15.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. Write a balanced equation first.

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



# Chapter 12 Stoichiometry Guided Reading Answers

**Robert R. Redfield**



**Chapter 12 Stoichiometry Guided Reading Answers:**

*Books in Print Supplement* ,2002

*Holt Chemistry* Ralph Thomas Myers,2004

The Software Encyclopedia ,1988

## Whispering the Techniques of Language: An Emotional Journey through **Chapter 12 Stoichiometry Guided Reading Answers**

In a digitally-driven world where monitors reign great and quick conversation drowns out the subtleties of language, the profound secrets and psychological nuances hidden within words usually get unheard. Yet, situated within the pages of **Chapter 12 Stoichiometry Guided Reading Answers** a charming fictional value pulsating with organic feelings, lies an exceptional journey waiting to be undertaken. Composed by a skilled wordsmith, this wonderful opus encourages visitors on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within the very material of every word. Within the psychological depths of this moving review, we will embark upon a heartfelt exploration of the book's key subjects, dissect their charming publishing design, and fail to the effective resonance it evokes heavy within the recesses of readers' hearts.

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