In the list below, circle the properties of acids and box the properties of bases.

Sour taste Feel slippery Contains hydroxide ions

Turns litmus blue Reacts with metals to form H Turns litmus red

Bitter taste Contains hydrogen ions Corrosive

Complete the chart below:

Acid-Base Definitions		
Type	Acid Definition	Base Definition
Arrhenius	Donates an H* ion in solution	Donates an OH ion in solution
Bronsted-Lowry	Donates an H* ion	Accepts an H* ion
Lewis	Accepts an electron pair	Donates an electron pair

- 3. What term(s) describe a substance that can act as an acid or a base? Amphoteric or amphiprotic
- 4. Neutralization reactions always produce a SALT and WATER
- If the [H<sup>\*</sup>] is greater than the [OH<sup>\*</sup>], the solution is acidic basic neutral
- 6. If the [OH] is greater than the [H], the solution is acidic basic neutral
- 7. If the [H'] is equal to the [OH'], the solution is acidic basic neutral
- 8. If the pH is 7, the solution is acidic basic neutral
- If the pH is greater than 7, the solution is acidic basic neutral
- 10. If the pH is less than 7, the solution is acidic basic neutral
- 11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?

$$LiOH + HClO_3 \rightarrow LiCl + H_2O$$

(19.2 g HClO<sub>2</sub>) x (1 mol HClO<sub>2</sub>/84 g HClO<sub>2</sub>) x (1 mol LiOH/1 mol HClO<sub>2</sub>) = 0.228 mol LiOH

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?

$$Ca(OH)_2 + 2 HNO_3 \rightarrow Ca(NO_3)_2 + 2 H_2O$$

(132.4 g HNO<sub>3</sub>) x (1 mol HNO<sub>3</sub>/63 g HNO<sub>3</sub>) x (1 mol Ca(OH)<sub>2</sub>/2 mol HNO<sub>3</sub>) x (74 g Ca(OH)<sub>2</sub>/1 mol Ca(OH)<sub>2</sub>) = 77.76 g Ca(OH)<sub>2</sub>

# **Chapter 19 Review Activity Acids And Bases Answers**

**Yicheng Fang** 

**Chapter 19 Review Activity Acids And Bases Answers:** 

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#### **Table of Contents Chapter 19 Review Activity Acids And Bases Answers**

- 1. Understanding the eBook Chapter 19 Review Activity Acids And Bases Answers
  - The Rise of Digital Reading Chapter 19 Review Activity Acids And Bases Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter 19 Review Activity Acids And Bases Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 19 Review Activity Acids And Bases Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter 19 Review Activity Acids And Bases Answers
  - Personalized Recommendations
  - Chapter 19 Review Activity Acids And Bases Answers User Reviews and Ratings
  - Chapter 19 Review Activity Acids And Bases Answers and Bestseller Lists

- 5. Accessing Chapter 19 Review Activity Acids And Bases Answers Free and Paid eBooks
  - o Chapter 19 Review Activity Acids And Bases Answers Public Domain eBooks
  - Chapter 19 Review Activity Acids And Bases Answers eBook Subscription Services
  - Chapter 19 Review Activity Acids And Bases Answers Budget-Friendly Options
- 6. Navigating Chapter 19 Review Activity Acids And Bases Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chapter 19 Review Activity Acids And Bases Answers Compatibility with Devices
  - Chapter 19 Review Activity Acids And Bases Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Chapter 19 Review Activity Acids And Bases Answers
  - Highlighting and Note-Taking Chapter 19 Review Activity Acids And Bases Answers
  - Interactive Elements Chapter 19 Review Activity Acids And Bases Answers
- 8. Staying Engaged with Chapter 19 Review Activity Acids And Bases Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 19 Review Activity Acids And Bases Answers
- 9. Balancing eBooks and Physical Books Chapter 19 Review Activity Acids And Bases Answers
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Chapter 19 Review Activity Acids And Bases Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 19 Review Activity Acids And Bases Answers
  - Setting Reading Goals Chapter 19 Review Activity Acids And Bases Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 19 Review Activity Acids And Bases Answers
  - Fact-Checking eBook Content of Chapter 19 Review Activity Acids And Bases Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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