

Chapter 3 Assessment

Vocabulary

Select the correct term to complete the sentences.

kilogram	gram	atom
element	compound	system
energy	Calorie	newton
heat	thermometer	Fahrenheit
Calorie	force	conservation of energy

Section 3.1

- The basic SI unit of mass, approximately equal to 1000 grains of rice, is the ____.
- A pure substance that cannot be broken down into other substances is known as a(n) ____.
- A combination of elements that most often has properties different than the elements from which it is made is called a(n) ____.
- The smallest particle of matter which retains the identity of the element it comes from is called a(n) ____.
- An amount of mass equal to one-thousandth of a kilogram is a(n) ____.

Section 3.2

- A small group of related things that work together may be called a ____.
- A quantity that measures the ability to cause change in a physical system in units of joules is known as ____.
- A unit of energy larger than a joule used to measure the energy available in food is the ____.
- An SI unit used to measure force, equal to less than one-quarter of a pound, is the ____.

- A natural law that says energy cannot be created or destroyed is the law of ____.
- An action, measured in newtons, that has the ability to transfer energy or change motion is called a(n) ____.
- A(n) ____ is an instrument used to measure temperature.
- The temperature scale on which the freezing point of water is 32 degrees and boiling point of water is 212 degrees is the ____ scale.
- The temperature scale on which the freezing point of water is 0 degrees and boiling point of water is 100 degrees is the ____ scale.
- When thermal energy flows from hot to cold it is called ____.

Concepts

Section 3.1

- Scientists find it useful to use both grams and kilograms for measuring mass. Why is it necessary to have two different SI units of mass?
- Mass describes:
 - an object's size.
 - the amount of matter in an object.
 - the type of elements in an object.
- What laboratory instrument is most often used to measure mass?
- Identify the following substances as an *element* or a *compound*:
 - ____ table salt (sodium chloride)
 - ____ oxygen gas
 - ____ rust
 - ____ iron

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**Ralph M. Feather, Susan Leach
Snyder, Dale T. Hesser**



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Chemistry John S. Phillips, Cheryl Wistrom, 2000 *Glencoe Mathematics* William Collins, 1999 **Lesson Plan Bklt**
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Table of Contents Chapter 16 Study Guide Answers Glencoe Physics

1. Understanding the eBook Chapter 16 Study Guide Answers Glencoe Physics
 - The Rise of Digital Reading Chapter 16 Study Guide Answers Glencoe Physics
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 16 Study Guide Answers Glencoe Physics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 16 Study Guide Answers Glencoe Physics
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 16 Study Guide Answers Glencoe Physics
 - Personalized Recommendations
 - Chapter 16 Study Guide Answers Glencoe Physics User Reviews and Ratings

- Chapter 16 Study Guide Answers Glencoe Physics and Bestseller Lists
- 5. Accessing Chapter 16 Study Guide Answers Glencoe Physics Free and Paid eBooks
 - Chapter 16 Study Guide Answers Glencoe Physics Public Domain eBooks
 - Chapter 16 Study Guide Answers Glencoe Physics eBook Subscription Services
 - Chapter 16 Study Guide Answers Glencoe Physics Budget-Friendly Options
- 6. Navigating Chapter 16 Study Guide Answers Glencoe Physics eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 16 Study Guide Answers Glencoe Physics Compatibility with Devices
 - Chapter 16 Study Guide Answers Glencoe Physics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 16 Study Guide Answers Glencoe Physics
 - Highlighting and Note-Taking Chapter 16 Study Guide Answers Glencoe Physics
 - Interactive Elements Chapter 16 Study Guide Answers Glencoe Physics
- 8. Staying Engaged with Chapter 16 Study Guide Answers Glencoe Physics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 16 Study Guide Answers Glencoe Physics
- 9. Balancing eBooks and Physical Books Chapter 16 Study Guide Answers Glencoe Physics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 16 Study Guide Answers Glencoe Physics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 16 Study Guide Answers Glencoe Physics
 - Setting Reading Goals Chapter 16 Study Guide Answers Glencoe Physics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 16 Study Guide Answers Glencoe Physics
 - Fact-Checking eBook Content of Chapter 16 Study Guide Answers Glencoe Physics
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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