(Class 11)

Question 13.13:

A gas in equilibrium has uniform density and pressure throughout its volume. This is strictly true only if there are no external influences. A gas column under gravity, for example, does not have uniform density (and pressure). As you might expect, its density decreases with height. The precise dependence is given by the so-called law of atmospheres $n_2 = n_1 \exp\left[-mg\left(h_2 - h_1\right)/kBT\right]$

Where n_2 , n_1 refer to number density at heights h2 and h1 respectively. Use this relation to derive the equation for sedimentation equilibrium of a suspension in a liquid column:

$$n_2 = n_1 \exp \left[-mg \, NA(\rho - P') \, (h_2 - h_1) / \, (\rho RT)\right]$$

Where ρ is the density of the suspended particle, and ρ' that of surrounding medium. [N_A is Avogadro's number, and R the universal gas constant.]

[Hint: Use Archimedes principle to find the apparent weight of the suspended particle.]

__Answer 13.13:

According to the law of atmospheres, we have: $n_2 = n_1 \exp \left[-mg \left(h_2 - h_1\right)/kBT\right]$...

 n_1 is the number density at height h_1 , and n_2 is the number density at height h_2 , m_3 is the weight of the particle suspended in the gas column.

Density of the medium =
$$\rho$$

Density of the suspended particle = ρ

Mass of one suspended particle = m'

Mass of the medium displaced = m

Volume of a suspended particle = V

According to Archimedes' principle for a particle suspended in a liquid column, the effective weight of the suspended particle is given as:

Weight of the medium displaced - Weight of the suspended particle

$$= mg - m'g$$

$$= mg - V \rho' g = mg - \left(\frac{m}{\rho}\right) \rho' g$$

$$= mg \left(1 - \frac{\rho'}{\rho}\right) \qquad -(ii)$$

Gas constant, R = kBN

$$k_{B} = \frac{R}{N}$$
 (III)

Substituting equation (ii) in place of mg in equation (i) and then using equation (iii), we get: $n_2 = n_1 \exp \left[-mg \left(h_2 - h_1\right)/kBT\right]$

$$= n_1 \exp \left[-mg\left(-\frac{\rho^*}{\rho}\right)_{(b_b-b_b)} \frac{N}{RT}\right]$$

$$= n_1 \exp \left[-mg\left(\rho - \rho^*\right)_{(b_b-b_b)} \frac{N}{RT\rho}\right]$$

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Chapter 13 Physics Problems Answers

A. Douglas Davis

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