HETEROGENEOUS AQUEOUS SYSTEMS

Section Review

Objectives

- Distinguish between a suspension and a solution
- Identify the distinguishing characteristic of a colloid.

Vocabulary

- suspension:
- colloid.
- Tyndall effect

- Brownian motion
- emulsion

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

than those of a solution. Gravity or ___ 2 __ will separate the suspended particles from a suspension. The particles in a colloid generally do not settle out under gravity, and they pass through ordinary filter paper unchanged. 3 are good at scattering light, as are suspensions, as evidenced by the ___4__. Colloidal dispersions also exhibit 5 motion. The particles in solutions 7. are small 6 or 7 . They cannot be trapped by filter paper, nor do they exhibit the Tyndall effect. 8 are colloidal dispersions of liquids in liquids.

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- 11. The scattering of light by colloidal particles is called Brownian motion.
- 12. Heterogeneous aqueous systems can be separated by filtration.

Emulsifying agents maintain the ___9 __ of an emulsion and allow the formation of __10 __ between liquids that do not ordinarily mix.

Chapter 15 Water And Aqueous Systems Worksheet Answers

B Lingard

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