

Study Question – Chapter 17 (Electrochemistry)

18A. What combination will yield the greatest E°_{cell} ?

$E^{\circ}_{\text{red}} / \text{V}$			
$\text{Cu}^{2+}(\text{aq}) + 2\text{e}^{-} \rightarrow \text{Cu}(\text{s})$	+0.34	$\text{In}^{3+}(\text{aq}) + 3\text{e}^{-} \rightarrow \text{In}(\text{s})$	-0.34
$\text{Cu}^{2+}(\text{aq}) + \text{e}^{-} \rightarrow \text{Cu}^{+}(\text{aq})$	+0.19	$\text{Cr}^{2+}(\text{aq}) + 2\text{e}^{-} \rightarrow \text{Cr}(\text{s})$	-0.91

- (A) $3\text{Cu}^{2+}(\text{aq}) + \text{In}^{3+}(\text{aq}) \rightarrow 3\text{Cu}^{+}(\text{aq}) + \text{In}(\text{s})$ (B) $3\text{Cu}^{2+}(\text{aq}) + 2\text{In}(\text{s}) \rightarrow 3\text{Cu}(\text{s}) + 2\text{In}^{3+}(\text{aq})$
 (C) $\text{Cu}^{2+}(\text{aq}) + \text{Cr}(\text{s}) \rightarrow \text{Cu}(\text{s}) + \text{Cr}^{2+}(\text{aq})$ (D) $2\text{Cu}^{2+}(\text{aq}) + \text{Cr}(\text{s}) \rightarrow 2\text{Cu}^{+}(\text{aq}) + \text{Cr}^{2+}(\text{aq})$

Work:

Practice Question – Chapter 17 (Electrochemistry)

18B. What is the strongest reducing agent?

$E^{\circ}_{\text{red}} / \text{V}$			
$\text{Cu}^{2+}(\text{aq}) + 2\text{e}^{-} \rightarrow \text{Cu}(\text{s})$	+0.34	$\text{Cr}^{3+}(\text{aq}) + 3\text{e}^{-} \rightarrow \text{Cr}(\text{s})$	-0.74

- (A) $\text{Cu}^{2+}(\text{aq})$ (B) $\text{Cr}^{3+}(\text{aq})$ (C) $\text{Cu}(\text{s})$ (D) $\text{Cr}(\text{s})$

Work:

Chapter 17 Electrochemistry Answers

Thomas Elliott Taylor



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Chemistry Workbook For Dummies Peter J. Mikulecky, Katherine Brutlag, Michelle Rose Gilman, Brian Peterson, 2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence Problem Exercises for General Chemistry George Gilbert Long, Forrest C. Hentz, 1982 New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions 3rd Edition Disha Experts, 2019-10-01 As NTA introduces Numeric Answer Questions in JEE Main Disha launches the Questions the 3rd latest updated edition of New Pattern NTA JEE Main Quick Guide in Chemistry with Numeric Answer Questions This study material is developed for quick revision and practice of the complete syllabus of the JEE Main Exam in a short span of 40 days The book can prove to be the ideal material for class 12 students as they can utilise this book to revise their preparation immediately after the board exams The book contains 27 chapters of class 11 12 and each Chapter contains JEE Main 6 Years at a Glance i e JEE Main 2019 2014 with TOPIC WISE Analysis Detailed Concept Maps covers entire JEE Syllabus for speedy revision IMPORTANT CRITICAL Points of the Chapter for last minute revision TIPS to PROBLEM SOLVING to help students to solve Problems in shortest possible time Exercise 1 CONCEPT BUILDER A Collection of Important Topic wise MCQs to Build Your Concepts Exercise 2 CONCEPT APPLICATOR A Collection of Quality MCQs that helps sharpens your concept application ability Exercise 3 Numeric Answer Questions A Collection of Quality Numeric Answer Questions as per the new pattern of JEE Answer Keys Detailed Solutions of all the Exercises and Past years problems are provided at the end of the chapter

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