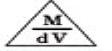


Density Calculations Worksheet I



Find the unknown quantity:

a)	d = 3 g/mL V = 100 mL
	M = ?
	m=dxv
	= (3glmL)(100mL)
	= 3 × 102 g

=?
=950 mL
= 95 g
=
$$\frac{m}{V}$$

= $\frac{159}{950mL}$ = 0.109 mL
= 0.109 mL
= $\frac{209}{0.591cm^3}$ = $\frac{209}{0.591cm^3}$ = $\frac{4 \times 10^3 cm^3}{0.591cm^3}$

Find the unknown quantity (CONVERT FIRST to g or mL)

WORD PROBLEMS

A block of aluminum occupies a volume of 15.0 mL and weighs 40.5 g. What is its density?

Mercury metal is poured into a graduated cylinder that holds exactly 22.5 mL. The mercury used to fill the cylinder weighs 306.0 g. From this information, calculate the density of mercury.

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RM Cervero

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