

EXERCISE NO. 4

DENSITY, SPECIFIC GRAVITY AND SPECIFIC VOLUME

I. INTENDED LEARNING OUTCOMES

At the end of the exercise, the students should be able to:

1. calculate density, specific gravity and specific volume of different liquids.
2. determine the specific gravity of unknown liquids using the data obtained from the specific gravity bottle (pycnometer) method.
3. calculate weight or volume of the substance given the specific gravity.

II. DISCUSSION

Density is defined as mass per unit volume of a substance at a fixed temperature and pressure. Density can be calculated by dividing the mass of a substance by its volume, expressed as: $Density = \frac{Mass}{Volume}$. The density of liquids is usually expressed as grams per milliliter (g/mL).

Specific gravity is the ratio of the weight of a given substance to the weight of an equal volume of water at a similar temperature, expressed as:

$$\text{Specific gravity} = \frac{\text{Weight of substance}}{\text{Weight of equal volume of water}}$$

In the U.S. Pharmacopeia, the standard temperature for specific gravities is 25°C, unless otherwise stated. Specific gravity is a dimensionless property and therefore has no units. The specific gravity of water is 1; therefore, the volume of water in milliliters is usually equal to its weight in grams. If the weight of a particular volume of a substance is known, its specific gravity can be calculated by dividing its weight by its volume:

$$\text{Specific gravity} = \frac{\text{Weight of substance (g)}}{\text{Volume of substance (mL)}}$$

This equation is similar to the one for density. So, an easy way to calculate specific gravity is to calculate the density and write the value without any units. If the specific gravity of a solution is known, interconversions between the volume of a substance and its weight are possible using the following expressions:

$$\text{Weight of substance (g)} = \text{Volume of substance (mL)} \times \text{Specific gravity}$$

$$\text{Volume of substance (mL)} = \frac{\text{Weight of substance (g)}}{\text{Specific gravity}}$$

Density And Specific Gravity Laboratory Manual

Arthur Herbert Church



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Table of Contents Density And Specific Gravity Laboratory Manual

1. Understanding the eBook Density And Specific Gravity Laboratory Manual
 - The Rise of Digital Reading Density And Specific Gravity Laboratory Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Density And Specific Gravity Laboratory Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Density And Specific Gravity Laboratory Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Density And Specific Gravity Laboratory Manual
 - Personalized Recommendations
 - Density And Specific Gravity Laboratory Manual User Reviews and Ratings
 - Density And Specific Gravity Laboratory Manual and Bestseller Lists

5. Accessing Density And Specific Gravity Laboratory Manual Free and Paid eBooks
 - Density And Specific Gravity Laboratory Manual Public Domain eBooks
 - Density And Specific Gravity Laboratory Manual eBook Subscription Services
 - Density And Specific Gravity Laboratory Manual Budget-Friendly Options
6. Navigating Density And Specific Gravity Laboratory Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Density And Specific Gravity Laboratory Manual Compatibility with Devices
 - Density And Specific Gravity Laboratory Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Density And Specific Gravity Laboratory Manual
 - Highlighting and Note-Taking Density And Specific Gravity Laboratory Manual
 - Interactive Elements Density And Specific Gravity Laboratory Manual
8. Staying Engaged with Density And Specific Gravity Laboratory Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Density And Specific Gravity Laboratory Manual
9. Balancing eBooks and Physical Books Density And Specific Gravity Laboratory Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Density And Specific Gravity Laboratory Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Density And Specific Gravity Laboratory Manual
 - Setting Reading Goals Density And Specific Gravity Laboratory Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Density And Specific Gravity Laboratory Manual
 - Fact-Checking eBook Content of Density And Specific Gravity Laboratory Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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