

Balancing Equations & Types of Reactions WS 1

Practice Problems

In your notebook, solve the following problems.

NOTE: A skeleton equation is an unbalanced equation.

SECTION 10.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.



2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.



3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.



4. Write a balanced equation for the production of HCl gas from its elements.



5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



Hydrochloric acid and calcium carbonate react to produce carbon dioxide, calcium chloride, and water.

6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



2 solid silver reacts w/ 1 solid sulfur to create solid silver sulfide.

Describing Chemical Reactions Section 1 Review Answers

Michael J. Padilla



Describing Chemical Reactions Section 1 Review Answers:

Essentials of Chemical Reaction Engineering H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Holt Chemistry** Ralph Thomas Myers, 2004 **Glencoe Science** McGraw-Hill Staff, 2001-06 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement *Science in Your World: Teacher edition* Jay K. Hackett, 1991

Designs for Science Literacy American Association for the Advancement of Science, 2001-03-22 The call for science curriculum reform has been made over and over again for much of the twentieth century Arguments have been made that the content of the curriculum is not appropriate for meeting the individual and social needs of people living in the modern world that the curriculum has become overstuffed with topics and does not serve students especially well and above all that the curriculum does not generate the student learning it is expected to produce The latest volume in a continuing series of publications from the AAAS designed to reform science education *Designs for Science Literacy* presupposes that curriculum reform must be considerably more extensive and fundamental than the tinkering with individual courses and subjects that has been going on for decades *Designs* deals with the critical issues involved in assembling sound instructional materials into a new coherent K 12 whole The book pays special attention to the need to link science oriented studies to the arts and humanities and also proposes how to align the curriculum with an established set of learning goals while preserving the American tradition of local responsibility for the curriculum itself If fundamental curriculum reform is ever to occur a new process for creating alternatives will have to be developed *Designs for Science Literacy* provides the groundwork for such a process *The Chemical Review*, 1883 *Introductory Algebra* Arnold R. Steffensen, Lee Murphy Johnson, 1994

Technical Section Proceedings Canadian Pulp and Paper Association. Technical Section, 1982 Annual meeting held after the end of the calendar year covered by the proceedings *Earth Science*, 2001 *Study Guide, Chemical Principles, Fifth Edition*, Zumdahl Paul B. Kelter, Steven S. Zumdahl, 2004-04 **Focus on Earth Science**, 2001 **Glencoe Physical iScience, Grade 8, Reading Essentials, Student Edition** McGraw Hill, 2004-08-02 Reading Essentials student edition provides an interactive reading experience to improve student comprehension of science content It makes lesson content

more accessible to struggling students and supports goals for differentiated instruction Students can highlight text and take notes right in the book **Chemical Molecular Science** Conrad L. Stanitski,2004-08 Science Voyages Glencoe,1999-05

Computational Medicinal Chemistry for Drug Discovery Patrick Bultinck,Hans De Winter,Wilfried Langenaeker,Jan P. Tollenare,2003-12-17 Observing computational chemistry s proven value to the introduction of new medicines this reference offers the techniques most frequently utilized by industry and academia for ligand design Featuring contributions from more than fifty pre eminent scientists Computational Medicinal Chemistry for Drug Discovery surveys molecular structure computation intermolecular behavior ligand receptor interaction and modeling responding to market demands in its selection and authoritative treatment of topics The book examines molecular mechanics semi empirical methods wave function based quantum chemistry density functional theory 3 D structure generation and hybrid methods **Te HS&T J**
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