CHEMISTRY HONORS DESIGN A HANDWARMER LAB ACTIVITY

Prior to 4/3

- 1. Copy Onestion
- 2. Read purpose
- 3. Create Safety Table for every Chemical you'll be using in the lab (Example at bottom of page)
 - a. NaOH e. NaCH₂CO₂
 - b. NH,NO, f. Na₂CO, c. CaCl₂ g. NH,Cl
 - A. Lici
- 4. Copy Cost of Chemicals

Solid	NH ₄ NO ₃	CaCl ₂	LiCI	NaCH ₂ CO ₂	Ma ₂ CO ₃	NH ₄ Cl
Cost	19.50	11.25	68.30	27.30	8.05	12.11
CSN/RC or						

(\$)/Kg

 Copy Pre Lab Procedure into Lab book in a procedure/observation t chart (paraphrase the procedure. You do not need to write it word for word)

The links of the SDS is found below

https://fscimage.fishersci.com/msds/21300.htm

http://archo.dfs.lps.org/Chemicals/Ammonium%20Nitrate.pdf

https://www.flinnsci.com/globalassets/flinn-scientific/all-free-pdfs/sds196_calcium-chloride.pdf/

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RTIFD+100G&vendorId=VN00033897&countryCode=US&language=en

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DESIGN A HAND WARMER LAB

Question: Which chemical should be used in the making of a hand warmer?

Purpose:

Your team is in charge of designing a hand warmer with the following criteria:

- 1. The hand warmer must contain 10 g of an ionic solid and an inner pouch filled with 40 mL of water.
- Activating the hand warmer must increase the temperature of the resulting solution by at least 20 °C
- Cost effectiveness and safety

Safety Precautions

Prior to experiment, a basic understanding of chemical hazards is needed. In general, avoid contact of all chemicals with eyes and skin. Wear chemical splash goggles. Wash hands thoroughly with soap and water before leaving the laboratory.

Pre Lab Procedure

- A. Working in pairs, determine the heat energy change for the dissolving of NaOH (ΔH...)
 - Determine the safety hazard for NaOH. Record the safety hazard in your lab book.

Designing A Hand Warmer Pre Lab Answers

Rolf F. Rehe

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- 1. Understanding the eBook Designing A Hand Warmer Pre Lab Answers
 - The Rise of Digital Reading Designing A Hand Warmer Pre Lab Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Designing A Hand Warmer Pre Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Designing A Hand Warmer Pre Lab Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Designing A Hand Warmer Pre Lab Answers
 - Personalized Recommendations
 - Designing A Hand Warmer Pre Lab Answers User Reviews and Ratings
 - Designing A Hand Warmer Pre Lab Answers and Bestseller Lists
- 5. Accessing Designing A Hand Warmer Pre Lab Answers Free and Paid eBooks

- Designing A Hand Warmer Pre Lab Answers Public Domain eBooks
- Designing A Hand Warmer Pre Lab Answers eBook Subscription Services
- Designing A Hand Warmer Pre Lab Answers Budget-Friendly Options
- 6. Navigating Designing A Hand Warmer Pre Lab Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Designing A Hand Warmer Pre Lab Answers Compatibility with Devices
 - Designing A Hand Warmer Pre Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Designing A Hand Warmer Pre Lab Answers
 - Highlighting and Note-Taking Designing A Hand Warmer Pre Lab Answers
 - Interactive Elements Designing A Hand Warmer Pre Lab Answers
- 8. Staying Engaged with Designing A Hand Warmer Pre Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Designing A Hand Warmer Pre Lab Answers
- 9. Balancing eBooks and Physical Books Designing A Hand Warmer Pre Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Designing A Hand Warmer Pre Lab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Designing A Hand Warmer Pre Lab Answers
 - Setting Reading Goals Designing A Hand Warmer Pre Lab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Designing A Hand Warmer Pre Lab Answers
 - Fact-Checking eBook Content of Designing A Hand Warmer Pre Lab Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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