

Determining Empirical Formulas

1) 75% carbon, 25% hydrogen

Answer:

 CH_4

Work:

$$\text{C } 75\% \rightarrow 75\text{g C} \left| \frac{1 \text{ mol}}{12.01\text{g C}} \right| = 6.24 \text{ mol} / 6.24 = 1$$

$$\text{H } 25\% \rightarrow 25\text{g H} \left| \frac{1 \text{ mol H}}{1.01\text{g H}} \right| = 24.75 \text{ mol} / 6.24 = 4$$



2) 52.7% potassium, 47.3% chlorine

Answer:

 KCl

Work:

$$\text{K } 52.7\% \rightarrow 52.7\text{g K} \left| \frac{1 \text{ mol K}}{39.10\text{g K}} \right| = 1.34 \text{ mol} / 1.33 = 1$$

$$\text{Cl } 47.3\% \rightarrow 47.3\text{g Cl} \left| \frac{1 \text{ mol Cl}}{35.45\text{g Cl}} \right| = 1.33 \text{ mol} / 1.33 = 1$$



3) 22.1% aluminum, 25.4% phosphorous, 52.5 % oxygen

Answer:

 AlPO_4

Work:

$$\text{Al } 22.1\% \rightarrow 22.1\text{g Al} \left| \frac{1 \text{ mol Al}}{26.98\text{g Al}} \right| = 0.819 \text{ mol} / 0.819 = 1$$

$$\text{P } 25.4\% \rightarrow 25.4\text{g P} \left| \frac{1 \text{ mol P}}{30.97\text{g P}} \right| = 0.820 \text{ mol} / 0.819 = 1$$

$$\text{O } 52.5\% \rightarrow 52.5\text{g O} \left| \frac{1 \text{ mol O}}{16.00\text{g O}} \right| = 3.28 \text{ mol} / 0.819 = 4$$



4) 13% magnesium, 87% bromine

Answer:

 MgBr_2

Work:

$$13\text{g Mg} \left| \frac{1 \text{ mol Mg}}{24.31\text{g Mg}} \right| = 0.54 \text{ mol Mg} / 0.54 = 1$$

$$87\text{g Br} \left| \frac{1 \text{ mol Br}}{79.90\text{g Br}} \right| = 1.09 \text{ mol Br} / 0.54 = 2$$



5) 32.4% sodium, 22.5% sulfur, 45.1% oxygen

Answer:

 Na_2SO_4

Work:

$$32.4\text{g Na} \left| \frac{1 \text{ mol}}{22.99\text{g Na}} \right| = 1.41 \text{ mol Na} / 0.70 = 2$$

$$22.5\text{g S} \left| \frac{1 \text{ mol S}}{32.37\text{g S}} \right| = 0.70 \text{ mol S} / 0.70 = 1$$

$$45.1\text{g O} \left| \frac{1 \text{ mol O}}{16\text{g O}} \right| = 2.82 \text{ mol O} / 0.70 = 4$$



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Determining Empirical Formulas Page 30 Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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