

Experiment 7
The Solution is Dilution
REPORT

1. Fill in this table using the instructions in the lab manual and from video:

Test Tube #	mL Stock Solution	mL DI water	Total Volume of solution (mL)	Molarity of solution, M
1	10.0	0.0		0.20
3	8.0	2.0		0.16
3	6.0	4.0		0.12
4	4.0	6.0		0.08
5	2.0	8.0		0.04

2. Complete the "Student analysis of colorimeter data and unknown" page of this experiment.

What is the concentration of the unknown solution of nickel(II) chloride? 0.143

Absorbance in video = 0.425

a = εcl

Questions needing solutions:

1. Does the absorbance of a solution depend on the concentration? yes
 (Explain your reasoning.)

2. Use the lab manual to find the wavelength of light absorbed and the color wheel (previous experiment) to fill in the table below.

Solution	Color of light absorbed	Color of solution
$\text{Co}(\text{NO}_3)_2(\text{aq})$		
$\text{NiCl}_2(\text{aq})$		

What is the relationship between the color of light absorbed by the solution and the color of the solution? (Explain below)

3. Suppose 25.00 g of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ (sucrose) is added to a 50.00 mL volumetric flask. DI water is added to the flask until the total volume of the solution is 50.00 mL. What is the molarity of the resulting solution? answer: _____

4. If 25.00 mL of the resulting solution is diluted with water to a total volume of 150 mL. What is the concentration of the diluted solution? answer: _____
 (Show calculation here)

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**Central Experimental Farm (Ottawa,
Ont.). Bacteriology Division**



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157504 9 Volume 2 Practical Chemistry Transition Metals ISBN 978 3 11 157384 7 *Nielsen's Food Analysis Laboratory Manual* B. Pam Ismail, S. Suzanne Nielsen, 2024-06-20 This fourth edition laboratory manual was written to accompany Nielsen's Food Analysis Sixth Edition by the same authors New to this fourth edition of the laboratory manual are three new chapters that complement both the textbook chapters and the laboratory exercises The book again contains four introductory chapters that help prepare students for doing food analysis laboratory exercises The 26 laboratory exercises in the manual cover 24 of the 35 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic Most of the laboratory exercises include the following background reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis Annual Report of the Commissioner of Health of the Commonwealth of Pennsylvania Pennsylvania. Commissioner of Health, Pennsylvania. Department of Health, 1908 Chemical Abstracts , 1909 *ICES Cooperative Research Report* , 1998 , *Laboratory Procedures for the Medical Office* Tom Palko, Hilda Palko, 1995 **The Ohio Public Health Journal of the Ohio State Board of Health** Ohio. State Board of Health, 1918 **Ohio's health. v. 9, 1918** , 1918

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