

# HOW IS THE FROG DIGESTIVE SYSTEM DIFFERENT FROM HUMANS

## FROG DIGESTIVE SYSTEM

Frogs do not have strong teeth. They use their teeth to hold the prey.

Frogs swallow the prey without chewing.

During deglutition, frogs blink or close their eyes.

Frogs have two sets of teeth: maxillary teeth and vomerine teeth.

The tongue of frog is attached to the starting point of the mouth.

The tip of tongue is folded backwards.

The tongue is very sticky.

Frogs have a shorter small intestine. The two parts of the small intestine of frogs are duodenum and ileum.

The duodenum is involved in the digestion of carbohydrates, proteins, lipids, and nucleic acids.

The absorption of nutrients occurs in the ileum.

The elimination of undigested materials occur through the cloaca.

The urinary bladder also opens into the cloaca since frogs do not have a urethra.

Frogs do not have an appendix.

## HUMAN DIGESTIVE SYSTEM

Humans have strong teeth. They use teeth to chew their food.

Chewing is a type of mechanical digestion in humans.

This is not seen in humans

Humans have one set of teeth in their oral cavity.

The tongue of humans is attached to the back of the mouth.

The tip of the human tongue is straight.

Human tongue is not sticky.

Humans have a long small intestine. It has three parts: duodenum, jejunum, and ileum.

Chemical digestion occurs in jejunum.

Absorption mainly occurs in the jejunum.

The elimination of undigested materials occur through rectum.

Humans have a separate urethra.

Humans have an appendix.

# Difference Between A Frog And Human Digestive System

**Leonard Landois**



## **Difference Between A Frog And Human Digestive System:**

Human Biology Activities Kit John R. Roland,1993-08-05 This collection of over 200 classroom tested activities and reproducible worksheets for students in grades 7 through 12 covers vital concepts in human biology and health including extensive coverage of AIDS These high interest lessons and worksheets get students actively involved in learning even students who are poorly motivated learning disabled or who lack English proficiency The lessons are written so you can easily accommodate your students various learning styles whether it s visual auditory and tactile Each lesson helps students make connections between new material and concepts they re already familiar with The book features 11 units covering all the body s systems such as circulatory digestive and immune systems and offers a detailed look at cells bones muscles and more Each unit provides enjoyable hands on activities that engage secondary students from building a cell model and testing foods for carbohydrates to dissecting a frog and making an action cartoon of a macrophage battling a microorganism For convenience the lessons are printed in a big spiral bound format that folds flat for photocopying **The Human Digestive**

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe} + \text{Co} \rightarrow \text{Fe} + \text{Co}^{2+}$  oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$  24 3  $\text{Ag} + \text{Ni} \rightarrow \text{Ag} + \text{Ni}^{2+}$  oxidation reduction 25  $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}^{2+}$  oxidation reduction 26  $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a  $\text{I}_2 + 5\text{CO} \rightarrow 2\text{CO}_2 + \text{I}_2$  b  $\text{Fe} + \text{Cu}^{2+} \rightarrow \text{Fe}^{2+} + \text{Cu}$  c  $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$  d  $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation

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reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^2$  oxidized reducing agent  $\text{O}^0$  to  $\text{O}^2$  reduced ox ag  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$   $\text{Li}^0$  to  $\text{Li}^1$  oxidized red ag  $\text{S}^0$  to  $\text{S}^2$  reduced ox ag  $3\text{Cs} + \text{Br}_2$