

## Physical and Chemical Changes and Properties of Matter Worksheet

Classify the following as chemical change (cc), chemical property (cp), physical change (pc), or physical property (pp).

- |                                     |                               |
|-------------------------------------|-------------------------------|
| 1. <u>PP</u> Heat conductivity      | 8. <u>CP</u> Combustible      |
| 2. <u>CC</u> Silver tarnishing      | 9. <u>PC</u> Water freezing   |
| 3. <u>PC</u> sublimation            | 10. <u>CC</u> Wood burning    |
| 4. <u>PC</u> magnetizing steel      | 11. <u>CP</u> Acid resistance |
| 5. <u>PP</u> length of metal object | 12. <u>PP</u> Brittleness     |
| 6. <u>PC</u> shortening melting     | 13. <u>CC</u> Milk souring    |
| 7. <u>CC</u> exploding dynamite     | 14. <u>CC</u> baking bread    |

Identify the following as being true or false to the left of the sentence.

- T 15. A change in size or shape is a physical change.  
T 16. A chemical change means a new substance with new properties was formed.  
F 17. An example of a chemical change is when water freezes.  
T 18. When platinum is heated, then cooled to its original state, we say this is a physical change.  
F 19. When milk turns sour, this is a physical change because a change in odor does not indicate a chemical change.  
T 20. When citric acid and baking soda mix, carbon dioxide is produced and the temperature decreases. This must be a chemical change.

Identify each of the following as a physical or chemical change.

21. CC You leave your bicycle out in the rain and it rusts.  
 22. PC A sugar cube dissolves.  
 23. CC Scientist break-up water into oxygen and hydrogen gas.  
 24. CC Burning coal for a barbecue.  
 25. PC Trimming a bush because it has grown too tall.

## Classifying Matter Worksheet

Classify each of the following substances as an element, a compound, a solution (homogenous mixture), or a heterogeneous mixture.

- |                                  |   |                                   |
|----------------------------------|---|-----------------------------------|
| 1. Sand<br>heterogeneous mixture | 2. Salt<br>Compound                       | 3. Pure Water<br>compound         |
| 4. Soil<br>heterogeneous mixture | 5. Soda just opened<br>homogenous mixture | 6. Pure air<br>homogenous mixture |

# Chapter Supplemental Problems Matter Properties And Changes Answer Key

**R Barnett**



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Chemistry McGraw-Hill Staff,2001-03     *Ebook: Chemistry: The Molecular Nature of Matter and Change*  
Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change     **Supplemental Problems -**  
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basis     *Elementary Differential Equations and Boundary Value Problems* William E. Boyce,Richard C. DiPrima,Douglas B.  
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the viewpoint of the applied mathematician whose interest in differential equations may sometimes be quite theoretical  
sometimes intensely practical and often somewhere in between The authors have sought to combine a sound and accurate  
but not abstract exposition of the elementary theory of differential equations with considerable material on methods of  
solution analysis and approximation that have proved useful in a wide variety of applications While the general structure of  
the book remains unchanged some notable changes have been made to improve the clarity and readability of basic material  
about differential equations and their applications In addition to expanded explanations the 11th edition includes new  
problems updated figures and examples to help motivate students The program is primarily intended for undergraduate  
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second year of study The main prerequisite for engaging with the program is a working knowledge of calculus gained from a  
normal two or three semester course sequence or its equivalent Some familiarity with matrices will also be helpful in the  
chapters on systems of differential equations     *Books in Print Supplement* ,2002     Methods for Euclidean Geometry  
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topics in mathematics Instead of carefully building geometries from axiom sets this book uses a wealth of methods to solve  
problems in Euclidean geometry Many of these methods arose where existing techniques proved inadequate In several cases  
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a geometry textbook but studying geometry in this way will also develop students appreciation of the subject and of  
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explores the application of a broad range of mathematical topics to the solution of Euclidean problems     Elementary

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