

CHAPTER 5: ELECTRONS IN ATOMS

CHEMISTRY

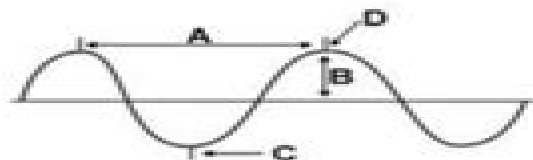
True/False

Indicate whether the statement is true or false.

- _____ 1. There are a total of 5 *d*-orbitals.
- _____ 2. The photoelectric effect describes the emission of electrons from a metal's surface when light at a specific frequency shines on it.
- _____ 3. Although the precise location of an electron cannot be determined, it is possible to determine the probability that an electron will occupy a certain region around the nucleus.
- _____ 4. In the visible light spectrum, red light has a greater frequency (energy) than violet light.
- _____ 5. $[\text{Kr}]5s^24d^{10}5p^1$ is the correct electron configuration for the element Bromine.
- _____ 6. The maximum number of allowed valence electrons in any atom is 8.
- _____ 7. Each element has its own unique emission spectrum.

Multiple Choice

Identify the choice that best completes the statement or answers the question.



- _____ 8. Which label identifies the amplitude of the wave shown?
a. A
b. B
c. C
d. D
- _____ 9. Which label identifies the wavelength of the wave shown?
a. A
b. B
c. C
d. D
- _____ 10. Which label identifies the trough of the wave shown?
a. A
b. B
c. C
d. D



- _____ 11. Which diagram shows a wave with the highest frequency?

Chapter Test Introduction To Atoms Answer Key

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Chapter Test Introduction To Atoms Answer Key:

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