

## Dilutions Worksheet

- 1) If I have 340 mL of a 1.5 M NaBr solution, what will the concentration be if I add 560 mL more water to it?
- 2) If I dilute 250 mL of 0.50 M lithium acetate solution to a volume of 750 mL, what will the concentration of this solution be?
- 3) If I leave 750 mL of 0.75 M sodium chloride solution uncovered on a windowsill and 150 mL of the solvent **evaporates**, what will the new concentration of the sodium chloride solution be?
- 4) To what volume would I need to add water to the evaporated solution in problem 3 to get a solution with a concentration of 0.50 M?
- 5) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?
- 6) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?
- 7) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

# Dilution Notes And Practice Answers

**Ronald A. McAllister**



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web non c est non zeilinger irene amazon com tr kitap Çerez tercihlerinizi seçin Çerez bildirimimizde ayrıntılı şekilde açıklandığı üzere alışveriş yapmanızı sağlamak alışveriş deneyiminizi iyileştirmek ve hizmetlerimizi sunmak için gerekli olan çerezleri ve benzer araçları kullanırız



