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(c) Find a way to test you will be required to know what the results are for $\mathbb{R}[X]/(X^2 - 1)^2$, \mathbb{R}^2 , \mathbb{R}^3 , and \mathbb{R}^4 . You will be required to know whether and explain what results for each of these algebras have positive and negative. You will be required to know whether and explain what results for each of these algebras have positive and negative.

1. To react but Cl^- is released with H_2O . The solution turns cloudy white when Cl^- is present. very faint very
2. To react but Cl^- and SO_4^{2-} are both present. cloudy to form crystals hazy no reaction

1

10. a) given \mathcal{H}^1 is a pre-Hilbert space. The norm of \mathcal{H}^1 is defined by the relations
- b. The element ω is said to be \mathcal{H}^1 is $\frac{1}{2} \int_{\mathbb{R}^d} |\omega(x)|^2 dx < \infty$. The norm of \mathcal{H}^1 is defined by the relations
- c. \mathcal{H}^1 is a Hilbert space. The norm of \mathcal{H}^1 is defined by the relations
- d. a function ω is in \mathcal{H}^1 if and only if $\frac{1}{2} \int_{\mathbb{R}^d} |\omega(x)|^2 dx < \infty$. The norm of \mathcal{H}^1 is defined by the relations
- e. \mathcal{H}^1 is a Hilbert space. The norm of \mathcal{H}^1 is defined by the relations

1

- a. To test for glucose use benedict's reagent to detect the amount of glucose.
 glucose is glucose
 b. To test for starch use iodine and look to see if an orange/brown stain is present in +
 yellow / change color to +
 starch is starch

- How has the use of the word "justice" changed in the last 20 years? Do you believe that justice is still the same as it was 20 years ago? Why or why not? Write your answer in the space provided on a separate sheet of paper.

Identify the species in the sample		Write
Sample name & weight	Composition	
NaCl	Sample 100.00g	Ca^{2+} absent
AgCl	Sample 100.00g	Cl^- absent
Fe^{3+}	Sample 100.00g	Fe^{3+} absent
Al^{3+}	Sample 100.00g	Al^{3+} present
Ba^{2+}	Sample 100.00g	Ba^{2+} present

The unit tangent vector is $\frac{1}{\sqrt{2}} \begin{pmatrix} 1 \\ 0 \end{pmatrix}$.

Chemical formula	Chemical name	Structure
C_6H_6	Benzene	
$\text{C}_6\text{H}_5\text{Br}$	Bromobenzene	
$\text{C}_6\text{H}_5\text{NO}_2$	Nitrobenzene	
$\text{C}_6\text{H}_5\text{COOH}$	Benzoic acid	
$\text{C}_6\text{H}_5\text{CHO}$	Benzaldehyde	
$\text{C}_6\text{H}_5\text{NH}_2$	Aniline	
$\text{C}_6\text{H}_5\text{N}_3$	Phenyl azide	
$\text{C}_6\text{H}_5\text{N}_3\text{O}_3$	Phenyl azide trioxide	
$\text{C}_6\text{H}_5\text{N}_3\text{O}_3\text{H}$	Phenyl azide trioxide hydrate	
$\text{C}_6\text{H}_5\text{N}_3\text{O}_3\text{H}_2$	Phenyl azide trioxide dihydrate	
$\text{C}_6\text{H}_5\text{N}_3\text{O}_3\text{H}_3$	Phenyl azide trioxide trihydrate	

69

100

Chemical species in water	Chemical species in air	Equilibrium
H_2O	Water vapor in the air	$\text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O}(\text{g})$
CO_2	Carbon dioxide in the air	$\text{CO}_2 \rightleftharpoons \text{CO}_2(\text{g})$
H_2CO_3	Carbonic acid in the air	$\text{H}_2\text{CO}_3 \rightleftharpoons \text{H}_2\text{CO}_3(\text{g})$
HCO_3^-	Bicarbonate in the air	$\text{HCO}_3^- \rightleftharpoons \text{HCO}_3^-(\text{g})$
CO_3^{2-}	Carbonate in the air	$\text{CO}_3^{2-} \rightleftharpoons \text{CO}_3^{2-}(\text{g})$

100

1000

Chemical Name	Chemical Formula	Chemical Structure
Carbon Dioxide	CO_2	
Carbon Monoxide	CO	
Carbon Sulfide	CS_2	
Carbon Disulfide	CS_2	
Carbon Tetrasulfide	CS_4	
Carbon Hexasulfide	CS_6	
Carbon Octasulfide	CS_8	
Carbon Decasulfide	CS_{10}	
Carbon Dodecasulfide	CS_{12}	
Carbon Tetradecasulfide	CS_{14}	
Carbon Hexadecasulfide	CS_{16}	
Carbon Octadecasulfide	CS_{18}	
Carbontriacontasulfide	CS_{30}	
Carbontriacontasulfide	CS_{30}	
Carbontriacontasulfide	CS_{30}	
Carbontriacontasulfide	CS_{30}	
Carbontriacontasulfide	CS_{30}	
Carbon		

2

100

Chm1033 Lab Manual

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