

Name _____ Date _____ Per _____

WORKSHEET: Chemical Bonding – Ionic & Covalent!

REMEMBER...

Ionic Bond

between a Metal and Non-Metal

(M + NM)

Covalent Bond

between a Non-Metal and Non-Metal

(NM + NM)

PART 1: Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type
NO ₂	N = non-metal	O = non-metal	covalent
NaCl	Na = Metal (M)	Cl = Nonmetal (NM)	I
SO ₂	S = NM	O = NM	C
PI ₃	P = NM	I = NM	C
MgBr ₂	Mg = M	Br = NM	I
CaO	Ca = M	O = NM	I
H ₂ O	H = NM	O = NM	C
K ₂ O	K = M	O = NM	I
AlF ₃	Al = M	F = NM	I
O ₂	O = NM	O = NM	C
CuCl ₂	Cu = M	Cl = NM	I
NO ₂	N = NM	O = NM	C
CO ₂	C = NM	O = NM	C
HF	H = NM	F = NM	C
Rb ₂ S	Rb = M	S = NM	I
NBr ₃	N = NM	Br = NM	C
Fe ₂ O ₃	Fe = M	O = NM	I
CCl ₄	C = NM	Cl = NM	C

Bonding Ionic Or Covalent Packet Answers

Michael DiSpezio



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