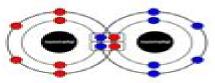
DRAWING COVALENT BONDS

Introduction:

A covalent bond forms between nonmetals. Covalent bonds involve sharing pairs of electrons between two atoms so that each atom gains a full outer valence shell. Sharing 1 pair of electrons creates a single covalent bond, sharing 2 pairs of electrons creates a double covalent bond, and sharing 3 pairs of electrons creates a triple covalent bond.



	How to Draw Simple Covalent Bond Diagrams						
1.	Draw the electron dot diagrams for each element. HINT: This will help you see the number of unpaired valence electrons for each element.						
2.	Count the total number of valence electrons between all atoms. HINT: The total number of valence electrons should be 8 in most cases, except 2 hydrogens. If there are not 8, add more atoms of the element with the least number of valence electrons to start.						
3.	Determine which element is the central atom in the molecule, if needed. HINT: The central atom is typically the atom with the largest valence number or the atom that there is only one of in the formula. If there are only 2 atoms in the formula, there is no central atom?						
4.	Pair up single electrons from different atoms to form covalent bonds and circle each pair. HIMT: Keep doing this until each single electron from all atoms have been paired up. You may need to use double or triple bonds to make this work. This step takes trial and error. If one arrangement is not working, try adding another element or trying a new structure. Practice makes perfect!						

Activity: Drawing Covalent Bond Diagrams:

- For each covalent compound, you have been given the element names and the chemical formula.
- Follow the steps listed above to create a covalent bonding diagram for the compound.
- Remember, this takes trial and error. The first 2 have been done for you as examples.

#	Nonmetal	Nonmetal	Formula	Bonding Diagram
	hydrogen	chlorine	HCI	
A.	H • (must share 1)	CI: (must share 1)	(Formula shows 1 H and 1 Cl)	H ©CI:
В.	hydrogen	oxygen	H ₂ O	
	H • (must share 1)	(must share 2)	(Formula shows 2 H and 1 O)	н⊜⊜∶

Covalent Bonding Assessment Answers

Robert M. Klein, George C. Enders

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- 1. Understanding the eBook Covalent Bonding Assessment Answers
 - The Rise of Digital Reading Covalent Bonding Assessment Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Covalent Bonding Assessment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bonding Assessment Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Covalent Bonding Assessment Answers
 - Personalized Recommendations
 - Covalent Bonding Assessment Answers User Reviews and Ratings
 - Covalent Bonding Assessment Answers and Bestseller Lists

- 5. Accessing Covalent Bonding Assessment Answers Free and Paid eBooks
 - Covalent Bonding Assessment Answers Public Domain eBooks
 - Covalent Bonding Assessment Answers eBook Subscription Services
 - Covalent Bonding Assessment Answers Budget-Friendly Options
- 6. Navigating Covalent Bonding Assessment Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Covalent Bonding Assessment Answers Compatibility with Devices
 - Covalent Bonding Assessment Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Bonding Assessment Answers
 - Highlighting and Note-Taking Covalent Bonding Assessment Answers
 - Interactive Elements Covalent Bonding Assessment Answers
- 8. Staying Engaged with Covalent Bonding Assessment Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - \circ Following Authors and Publishers Covalent Bonding Assessment Answers
- 9. Balancing eBooks and Physical Books Covalent Bonding Assessment Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bonding Assessment Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Covalent Bonding Assessment Answers
 - Setting Reading Goals Covalent Bonding Assessment Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Covalent Bonding Assessment Answers
 - Fact-Checking eBook Content of Covalent Bonding Assessment Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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