(v) When Y is heated with an alcoholic solution of potassium hydroxide, the alkene C, H, is formed. What type of reaction occurs to produce the alkene? (7)(b) Separate ethanolic solutions of X and Y were warmed with water containing dissolved silver nitrate. Describe what would be seen in each case, ignoring any differences in the rates of reaction. (2)Observation with X Observation with Y (c) The rates of hydrolysis of primary halogenoalkanes are affected by the strength of the bond between the carbon and the halogen atom. Is the C-Br bond weaker or stronger than the C-I bond? Explain why the bond strength differs. (2)

(Total for Question 22 = 12 marks)

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