



Name: \_\_\_\_\_  
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# COUNTING ATOMS

## WORKSHEET

- Chemical formulas can have three components.
  - The \_\_\_\_\_ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
  - The \_\_\_\_\_ which are numbers that are found on the lower \_\_\_\_\_ - hand side of each element symbol. This tells you the number of \_\_\_\_\_ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only \_\_\_\_\_ atom of this element in the molecule.
  - The \_\_\_\_\_ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be \_\_\_\_\_ by the \_\_\_\_\_ on the outside of the bracket found on the lower \_\_\_\_\_ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li <sub>2</sub> SO <sub>4</sub>	
Element	# of Atoms
Total	

7) H <sub>2</sub> COCH <sub>2</sub>	
Element	# of Atoms
Total	

10) NH <sub>4</sub> C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

13) Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	
Element	# of Atoms
Total	

2) HNO <sub>3</sub>	
Element	# of Atoms
Total	

5) NH <sub>4</sub> Cl	
Element	# of Atoms
Total	

8) Mg(OH) <sub>2</sub>	
Element	# of Atoms
Total	

11) NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

14) Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

3) U <sub>2</sub> O	
Element	# of Atoms
Total	

6) CaClO <sub>3</sub>	
Element	# of Atoms
Total	

9) Al(OH) <sub>3</sub>	
Element	# of Atoms
Total	

12) (NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub>	
Element	# of Atoms
Total	

15) Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

# Counting Atoms Study Guide Answers

**G. Marc Loudon, Joseph G. Stowell**



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

**Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

**Organic Chemistry Study Guide and Solutions Manual** Marc Loudon, Jim Parise, Lisa Bonner, 2024-08-09 The Study Guide and Solutions Manual offers several study tools for students Study Guide Links are brief supplements to text material and provide hints for study shortcuts approaches to problem solving or additional explanations of topics that typically cause difficulty Further Explorations are short discussions that provide additional depth on content in the text Reaction Reviews present the reactions in the chapter in detailed outline form This summary consolidates the reactions of the chapter in one place and presents not only the reactions themselves but also the essential features of their mechanisms and stereochemistry for cases in which these issues are discussed in the text Solutions are provided for all of the over 1600 problems in the text Many include worked out examples in which not only the answer is given but also the detailed approach to the solution is discussed

**Study Guide/Selected Solutions Manual** Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

Study Guide and Solutions Manual Neil E. Schore, 2002-08-02

*Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e* Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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edition contains over 1 800 problems many of them new and taken directly from the scientific literature The book is used at a wide variety of schools such as UC Berkeley Caltech Colorado Cornell Duke Harvard Illinois Maryland Purdue Yale Wisconsin and many more This edition provides students with more health examples drawn from modern medical practice as well as many cutting edge topics from modern synthetic organic chemistry In addition to the printed book students can rely on Sapling Learning's online homework platform for extra learning and assessment The platform offers automatic grading an easy to use interface and instructive feedback Instructors can select from a variety of existing problem sets over 1 000 of Loudon's problems are in the platform or they can modify the questions or author them from scratch Not only does the software allow students to easily draw and interact with structures it allows them to draw entire reaction mechanisms including showing the movement of electrons with curved electron arrows

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**Study Guide & Solutions Manual to Accompany Organic Chemistry, Third Edition** G. Marc Loudon, Joseph G. Stowell, 1995     *Chemistry - Study Guide to Accompany Moore/Davies/Collins* Clare T. Furse, 1978     **As You Like It in Plain and Simple English (A Modern Translation and the Original Version)** , 2011 Ever heard the phrase too much of a good thing That was actually coined by Shakespeare in this play Most people don't know it because when they hear the name Shakespeare they run and hide Let's face it if you don't understand Shakespeare then you are not alone If you have struggled in the past reading Shakespeare then BookCaps can help you out This book is a modern translation of As You Like It The original text is also presented in the book along with a comparable version of the modern text As You Like It follows its heroine Rosalind as she flees persecution in her uncle's court accompanied by her cousin Celia and Touchstone the court jester to find safety and eventually love in the Forest of Arden The play features one of Shakespeare's most famous and oft quoted speeches All the world's a stage We all need

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