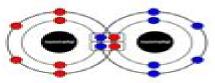
DRAWING COVALENT BONDS

Introduction:

A covalent bond forms between nonmetals. Covalent bonds involve sharing pairs of electrons between two atoms so that each atom gains a full outer valence shell. Sharing 1 pair of electrons creates a single covalent bond, sharing 2 pairs of electrons creates a double covalent bond, and sharing 3 pairs of electrons creates a triple covalent bond.



	How to Draw Simple Covalent Bond Diagrams						
1.	Draw the electron dot diagrams for each element. HINT: This will help you see the number of unpaired valence electrons for each element.						
2.	Count the total number of valence electrons between all atoms. HINT: The total number of valence electrons should be 8 in most cases, except 2 hydrogens. If there are not 8, add more atoms of the element with the least number of valence electrons to start.						
3.	Determine which element is the central atom in the molecule, if needed. HINT: The central atom is typically the atom with the largest valence number or the atom that there is only one of in the formula. If there are only 2 atoms in the formula, there is no central atom?						
4.	Pair up single electrons from different atoms to form covalent bonds and circle each pair. HIMT: Keep doing this until each single electron from all atoms have been paired up. You may need to use double or triple bonds to make this work. This step takes trial and error. If one arrangement is not working, try adding another element or trying a new structure. Practice makes perfect!						

Activity: Drawing Covalent Bond Diagrams:

- For each covalent compound, you have been given the element names and the chemical formula.
- Follow the steps listed above to create a covalent bonding diagram for the compound.
- Remember, this takes trial and error. The first 2 have been done for you as examples.

#	Nonmetal	Nonmetal	Formula	Bonding Diagram
	hydrogen	chlorine	HCI	
A.	H • (must share 1)	CI: (must share 1)	(Formula shows 1 H and 1 Cl)	H ©CI:
В.	hydrogen	oxygen	H ₂ O	
	H • (must share 1)	(must share 2)	(Formula shows 2 H and 1 O)	н⊜⊜∶

Covalent Bonding Workbook Answers

Nichole Vivion

Covalent Bonding Workbook Answers:

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The book delves into Covalent Bonding Workbook Answers. Covalent Bonding Workbook Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Covalent Bonding Workbook Answers, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Covalent Bonding Workbook Answers
 - Chapter 3: Covalent Bonding Workbook Answers in Everyday Life
 - Chapter 4: Covalent Bonding Workbook Answers in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 5. In chapter 4, this book will scrutinize the relevance of Covalent Bonding Workbook Answers in specific contexts. This chapter will explore how Covalent Bonding Workbook Answers is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Covalent Bonding Workbook Answers. The final chapter will summarize the key points that have been discussed throughout the book.
 - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Covalent Bonding Workbook Answers.

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