

# Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON-METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.)  $H_2$  hydrogen is diatomic



Single Bond

2.)  $F_2$  fluorine is diatomic



3.)  $O_2$  oxygen is diatomic



Double Bond

4.)  $N_2$  nitrogen is diatomic. Is this a triple bond?



5.)  $BF_3$  you need 3 fluorine atoms here



6.) Ammonia  $NH_3$  hint: how many hydrogen atoms are needed?



7.) Carbon dioxide  $CO_2$



8.) Methane  $CH_4$  careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.)  $SO_2$  hint: one pair of electrons from sulfur must be slit up for this one to work.



# Covalent Bonding Worksheet Answer Key

**Rachel Turoscy**



## Covalent Bonding Worksheet Answer Key:

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## Covalent Bonding Worksheet Answer Key Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Covalent Bonding Worksheet Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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