

Question Number	Acceptable Answers	Reject	Mark
22(b)	<p>X Cream precipitate Allow off white / creamy white / white-cream / (very) pale yellow (1)</p> <p>Y yellow precipitate (1)</p> <p>One mark for two correct colours but not precipitates Mention of precipitate without colours doesn't score Ignore identity of precipitates even if incorrect</p>	<p>Yellow/ creamy yellow precipitate for X</p> <p>Pale yellow precipitate for Y</p>	2

Question Number	Acceptable Answers	Reject	Mark
22(c)	<p>C-Br stronger / C-I weaker with an attempt at an explanation (correct or incorrect) (1)</p> <p>as bond is shorter/ Br (atom) is smaller / as nuclei are closer</p> <p>OR reverse argument (1)</p> <p>Allow Br is more electronegative/ there is a bigger electronegativity difference / bond is more polar / $C^{\delta+}$ and $Br^{\delta-}$ attract more strongly / Br is less shielded</p> <p>Second mark depends on first</p>	<p>Br_2 is smaller</p> <p>References to Br^- implying bond is ionic.</p> <p>Br is more reactive</p>	2

Total for Q22 = 12 marks

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Susie Jameson

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