

Question Number	Acceptable Answers	Mark
18(c)(i)	<p>IGNORE sf except 1</p> <p>If answer is 3.84 (mol dm⁻³), award 3 marks If not, number of moles of acid = $\frac{24.0 \times 4}{1000} = 0.096$ (1)</p> <p>EITHER number of moles ammonia = 0.096 in 25 cm³ (1)</p> <p>concentration of ammonia = $\frac{0.096 \times 1000}{25}$ = 3.84 (mol dm⁻³) (1)</p> <p>OR number of moles ammonia = 0.288 in 75 cm³ (1)</p> <p>concentration of ammonia = $\frac{0.288 \times 1000}{75}$ = 3.84 (mol dm⁻³) (1)</p> <p>IGNORE unit unless incorrect</p> <p>ALLOW ecf in both methods on their number of moles of ammonia</p>	3

Question Number	Acceptable Answers	Mark
18(c)(ii)	<p>IGNORE sf except 1 (concentration of ammonia in trichloromethane =) 0.16 (mol dm⁻³)</p> <p>ALLOW ecf from their answer to (c)(i), provided it is less than 4.0 and given to 2 or more sf</p>	1

Question Number	Acceptable Answers	Reject	Mark
18(c)(iii)	<p>Expression for K_c and answer needed for the mark $K_c = \frac{[\text{NH}_3(\text{aq})]}{[\text{NH}_3(\text{CHCl}_3)]}$ ALLOW one state symbol missing</p> <p>= $\frac{3.84}{0.16}$ = 24(.0)</p> <p>IGNORE sf, including 1 sf, and units ALLOW ecf from answers to (c)(i) and (c)(ii)</p>	<p>K_c expressions without both state symbols</p>	1

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Table of Contents Edexcel Ial Jan 2014 Chem Ms

1. Understanding the eBook Edexcel Ial Jan 2014 Chem Ms
 - The Rise of Digital Reading Edexcel Ial Jan 2014 Chem Ms
 - Advantages of eBooks Over Traditional Books
2. Identifying Edexcel Ial Jan 2014 Chem Ms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Edexcel Ial Jan 2014 Chem Ms
 - User-Friendly Interface
4. Exploring eBook Recommendations from Edexcel Ial Jan 2014 Chem Ms
 - Personalized Recommendations
 - Edexcel Ial Jan 2014 Chem Ms User Reviews and Ratings
 - Edexcel Ial Jan 2014 Chem Ms and Bestseller Lists
5. Accessing Edexcel Ial Jan 2014 Chem Ms Free and Paid eBooks

- Edexcel Ial Jan 2014 Chem Ms Public Domain eBooks
 - Edexcel Ial Jan 2014 Chem Ms eBook Subscription Services
 - Edexcel Ial Jan 2014 Chem Ms Budget-Friendly Options
6. Navigating Edexcel Ial Jan 2014 Chem Ms eBook Formats
- ePub, PDF, MOBI, and More
 - Edexcel Ial Jan 2014 Chem Ms Compatibility with Devices
 - Edexcel Ial Jan 2014 Chem Ms Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Edexcel Ial Jan 2014 Chem Ms
 - Highlighting and Note-Taking Edexcel Ial Jan 2014 Chem Ms
 - Interactive Elements Edexcel Ial Jan 2014 Chem Ms
8. Staying Engaged with Edexcel Ial Jan 2014 Chem Ms
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Edexcel Ial Jan 2014 Chem Ms
9. Balancing eBooks and Physical Books Edexcel Ial Jan 2014 Chem Ms
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Edexcel Ial Jan 2014 Chem Ms
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Edexcel Ial Jan 2014 Chem Ms
- Setting Reading Goals Edexcel Ial Jan 2014 Chem Ms
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Edexcel Ial Jan 2014 Chem Ms
- Fact-Checking eBook Content of Edexcel Ial Jan 2014 Chem Ms
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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