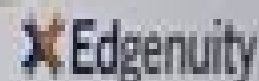


## Lab: Solubility Student Guide



### Pre-Lab Information

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<b>Purpose</b>	Explore the relationship between temperature and solubility using a laboratory procedure.
<b>Time</b>	Approximately 45 minutes
<b>Question</b>	What is the effect of temperature on the solubility of a solid in a liquid?
<b>Hypothesis</b>	If the temperature of the liquid is increased, then more sugar will dissolve because warm solutions hold more solute than cold solutions.
<b>Variables</b>	<i>Independent Variable:</i> temperature of water  <i>Dependent Variable:</i> amount of sugar that can be dissolved
<b>Summary</b>	You will measure the amount of sugar you can dissolve at four (or more) different temperatures of water and analyze the resulting data.

### Safety

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# Edgenuity Lab Solubility Answers

**Kōzō Shinoda**



### **Edgenuity Lab Solubility Answers:**

**Solutions and Solubilities** Michael R. J.. Dack,1975      *Handbook of Aqueous Solubility Data* Samuel H. Yalkowsky,Yan He,Parijat Jain,2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of      *Principles of Solution and Solubility* Kōzō Shinoda,1978      *Solubility data series* International Union of Pure and Applied Chemistry. Commission on Solubility Data,1979      **The Solubility of Nonelectrolytes** Joel Henry Hildebrand,Robert Lane Scott,1964

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