

Name _____

BONDING



Directions: Use the chart below and your periodic table to answer the questions that follow.

PART I: COMPARING ELEMENTS

Group One and Group Two both show different compounds made by bonding different elements together. There are two types of bonding in food science, as shown in these two groupings. Let's take a closer look at the elements in each compound to see what makes a compound IONIC or COVALENT.

IONIC BONDS	COVALENT BONDS
NaCl (Sodium Chloride)	H ₂ O (Dihydrogen Monoxide)
MgO (Magnesium Oxide)	CH ₄ (Methane)
Li ₂ O (Lithium Oxide)	CO ₂ (Carbon Dioxide)
KF (Potassium Fluoride)	HF (Hydrofluoric Acid)
FeBr ₃ (Iron III Bromide)	NH ₃ (Ammonia)
CaCl ₂ (Calcium Chloride)	NO ₂ (Nitrogen Dioxide)
NiI ₂ (Nickel Iodide)	C ₆ H ₁₂ O ₆ (Glucose)
BaS (Barium Sulfide)	CF ₄ (Carbon Tetrafluoride)

1. List out the FIRST element in each of the GROUP ONE compounds.
2. What do you notice about the FIRST element that makes up each compound in GROUP ONE?
3. What side of the periodic table do you find these FIRST elements on?
4. List out the SECOND element in each of the GROUP ONE compounds.
5. What do you notice about the SECOND element that makes up each compound in GROUP ONE?
6. What side of the periodic table do you find these SECOND elements on?

BIG IDEA #1: When a _____ and a _____ come together, they make an IONIC BOND.

7. What do you notice about ALL of the elements in GROUP TWO?
8. What side of the periodic table do you find ALL of these elements on?

Bonding Pogil 2 Covalent Bonds Answer Key

M.S. Sethi & M. Satake



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Table of Contents Bonding Pogil 2 Covalent Bonds Answer Key

1. Understanding the eBook Bonding Pogil 2 Covalent Bonds Answer Key
 - The Rise of Digital Reading Bonding Pogil 2 Covalent Bonds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Bonding Pogil 2 Covalent Bonds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Bonding Pogil 2 Covalent Bonds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Bonding Pogil 2 Covalent Bonds Answer Key
 - Personalized Recommendations

- Bonding Pogil 2 Covalent Bonds Answer Key User Reviews and Ratings
- Bonding Pogil 2 Covalent Bonds Answer Key and Bestseller Lists
- 5. Accessing Bonding Pogil 2 Covalent Bonds Answer Key Free and Paid eBooks
 - Bonding Pogil 2 Covalent Bonds Answer Key Public Domain eBooks
 - Bonding Pogil 2 Covalent Bonds Answer Key eBook Subscription Services
 - Bonding Pogil 2 Covalent Bonds Answer Key Budget-Friendly Options
- 6. Navigating Bonding Pogil 2 Covalent Bonds Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Bonding Pogil 2 Covalent Bonds Answer Key Compatibility with Devices
 - Bonding Pogil 2 Covalent Bonds Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Bonding Pogil 2 Covalent Bonds Answer Key
 - Highlighting and Note-Taking Bonding Pogil 2 Covalent Bonds Answer Key
 - Interactive Elements Bonding Pogil 2 Covalent Bonds Answer Key
- 8. Staying Engaged with Bonding Pogil 2 Covalent Bonds Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Bonding Pogil 2 Covalent Bonds Answer Key
- 9. Balancing eBooks and Physical Books Bonding Pogil 2 Covalent Bonds Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Bonding Pogil 2 Covalent Bonds Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Bonding Pogil 2 Covalent Bonds Answer Key
 - Setting Reading Goals Bonding Pogil 2 Covalent Bonds Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Bonding Pogil 2 Covalent Bonds Answer Key
 - Fact-Checking eBook Content of Bonding Pogil 2 Covalent Bonds Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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