

# CHEMICAL BONDING

## BASICS

### DOODLE NOTES

#### Bonding Basics

Small structure

Small structure

Small structure

Br

Three Main Types of Bonds

The type of bond is determined by the elements involved in the bond.

ionic bond

Transfer of electrons from one atom to another, resulting in oppositely charged ions that are held together by electrostatic forces.

Notes on:  
Three Main  
types of  
Chemical  
Bonds

Covale  
Bond

Formed between two non-metal atoms. The electrons are shared between the two atoms. Both atoms want the electrons, so they are held together by a strong force.

Metall  
Bond

Only forms between two metal atoms.

Electrons are shared between the two atoms.

Covalent Bond	Metallic Bond
<p>Two non-metal atoms share their valence electrons.</p> <p>Example: <math>\text{CO}_2</math></p>	<p>Only forms between two metal atoms.</p> <p>Example: <math>\text{Cu}</math></p>
<p>Two metal atoms share their valence electrons.</p> <p>Example: <math>\text{Cu}</math></p>	<p>Two non-metal atoms share their valence electrons.</p> <p>Example: <math>\text{NP}</math></p>

# Bonding Basics Guided Notes

**Jerry J. Weygandt, Paul D.  
Kimmel, Donald E. Kieso**



## **Bonding Basics Guided Notes:**

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