

Unit Test: Chemical Bonding (A) **ANSWER KEY**

100

Part I: Matching (2 points each) Please match the term with the correct definition. There will be extra choices and some choices may be used more than once.

- | | | |
|----------------------|-------------------------|----------------------|
| A. Electronegativity | F. Molecular Formula | K. Crystal |
| B. Covalent bonding | G. Linear | L. Pyramid |
| C. Ductile | H. Non-polar | M. Polar |
| D. Aqueous | I. Solid | N. VSEPR |
| E. Tetrahedral | J. Trigonal bipyramidal | O. Malleable |
| P. Octet Rule | Q. Ionic bonding | R. Empirical Formula |

1. K A 3-D, repeating pattern of + and - ions, formed by ionic compound
2. A Tendency for an atom to attract electrons when chemically bonded to another atom
3. B A sharing of electrons
4. P Atoms will gain or lose enough electrons in order to become isoelectronic with a noble gas
5. Q A transfer of electrons
6. R A chemical formula that is arranged in the smallest whole number ratio
7. D The term that means dissolved in water
8. F A chemical formula that describes the makeup of a single molecule
9. J The shape (geometry) that is always an exception to the octet rule
10. M A bond where electrons are shared unequally between atoms
11. L One of the shapes (geometries) that will always be polar
12. H A bond where electrons are shared equally between atoms

Part II: Multiple Choice (2 points each) Please circle the most correct answer. Write your answer on the given lines.

1. E A triple covalent bond shares _____ electrons between atoms.
a. two
b. three
c. four
d. five
e. six
f. Unable to tell
2. C Which of the following sets of atoms can be joined by a covalent bond?
a. Be and F
b. Na and Cl
c. S and F
d. Mg and Na
e. Li and Cl
f. None of these

Bonding Test Answer Key

Michael Hoelscher

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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reader needs to pass the Journeyman electrical competency exam About the Author Ray Holder has worked in the electrical industry for more than 40 years as an apprentice journeyman master field engineer estimator business manager contractor inspector and instructor He is a graduate of Texas State University and holds a Bachelor of Science Degree in Occupational Education A certified instructor of electrical trades he has been awarded a lifetime teaching certificate from the Texas Education Agency in the field of Vocational Education Mr Holder has taught thousands of students at Austin Community College Austin Texas Odessa College at Odessa Texas Technical Vocational Institute of Albuquerque New Mexico Howard College at San Angelo Texas and in the public school systems in Fort Worth and San Antonio Texas He is currently Director of Education for Electrical Seminars Inc of San Marcos Texas Mr Holder is an active member of the National Fire Protection Association International Association of Electrical Inspectors and the International Brotherhood of Electrical Workers

Structure, Bonding and Main Group Chemistry Rod Beavon, Alan Jarvis, 2003 The revised edition of the highly successful Nelson Advanced Science series for A Level Chemistry Structure Bonding and Main Group Chemistry provides full content coverage of Unit 1 of the AS and A2 specifications *CCEA GCSE Double Award Science* Denmour Boyd, Nora Henry, Frank McCauley, Alyn G. McFarland, James Napier, Roy White, 2017-12-18 Exam Board CCEA Level GCSE Subject Science First Teaching September 2017 First Exam June 2019 Build your students scientific thinking and practical skills with this textbook developed specifically for the 2017 GCSE specifications from the No 1 publisher for CCEA GCSE Science Develop understanding with clear Examples Tips and Practical activities Prepare students for assessment with Test Yourself questions Maths practice and Exam style questions throughout Supports Foundation and Higher tier students in one book

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