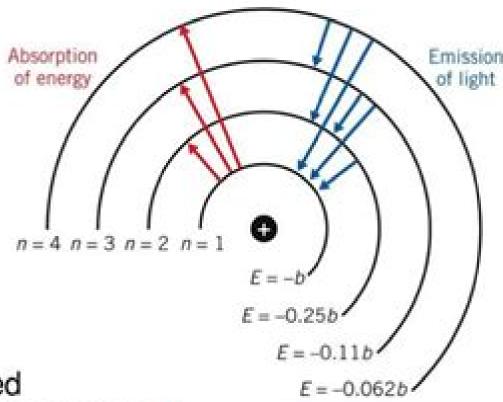
Bohr Model

—Energy Level Diagram for a Hydrogen Atom

- Absorption of photon
 - Electron raised to higher energy level
- Emission of photon
 - Electron falls to lower energy level



Energy levels are quantized

- Every time an electron drops from one energy level to a lower energy level
- Same frequency photon is emitted
- Yields line spectra

$$E = -\frac{b}{n^2} = -\frac{R_H hc}{n^2}$$

Bohr Model Of Hydrogen Gizmo Key

Helge Kragh

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include the historical background to the model Bohr s approach to his original derivation and corollary issues such as the role of angular momentum in the theory ionized helium the correspondence principle the fine structure constant de Broglie matter waves application of the theory to the diatomic hydrogen molecule and the magnetic field created by the orbiting electron It also includes student exercises a bibliography a list of important physical constants and a survey of Bohr s subsequent life and career Prov de l editor

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