

Lab Assessment and Review

This exercise is intended to assess basic lab skills and data handling practices required in Chem 185. To complete this review exercise, refer to the General Chemistry Notes if necessary. After the completion of the exercise, submit your work in Canvas for credit. If you have a lot of trouble with this exercise, discuss with your instructor for remedial steps.

Review Worksheets

Uncertainty and Significant Figures

Please read the discussions in your GCN (General Chemistry Notes) pages 80-83; 133-137 and answer the following questions.

- (400) 1. Every measurement carries uncertainty. The degree of uncertainty depends on the device used. When you record a measurement in lab, in addition to give the magnitude of the measurement, you must also convey the uncertainty associated with it. Observe each of the following devices from your drawer and determine its absolute uncertainty. How many decimal places should be used when recording the measurement with the device?

- Thermometer.
- 50-mL graduated cylinder.
- 1-mL measuring pipet.

2. Evaluate each of the following. Give the intermediate answer with a vertical dotted line to delineate the significant and non-significant digits. Write the final answer to the appropriate number of significant figures.

a. $21.901 - 13.21 + 4.0215 = 12.7115 = \boxed{12.71}$

b. $1.34 \times 10^5 + 10.1 \times 10^2 = 1.34 \times 10^5 + 0.0101 \times 10^5 = 1.3501 \times 10^5 = \boxed{1.35 \times 10^5}$

c. $(4.0 \times 10^{-6})(5.021 \times 10^{-3})(7.34993 \times 10^2) = 147615.9841 = 1.476159841 \times 10^5 = \boxed{1.5 \times 10^5}$

d. $(2.04 \times 10^{-6})^2 / (0.01231 \times 10^{-7}) = \frac{(4.1616 \times 10^{-12})}{(0.01231 \times 10^{-7})} = 3.38106 \times 10^4 = \boxed{3.38 \times 10^4}$

e. average of 123.45 and 1.4 = $\frac{(123.45 + 1.4)}{2} = 62.425 = \boxed{62.4}$

f. $(1.34 \times 10^2 + 3.2 \times 10^1) / (3.32 \times 10^{-6}) = \frac{(1.34 \times 10^2 + 0.32 \times 10^2)}{3.32 \times 10^{-6}} = \frac{1.66 \times 10^2}{3.32 \times 10^{-6}} = 5.00 \times 10^7 = \boxed{5.00 \times 10^7}$

Chspter 7 Section Assessment Answers Chemistry

Arturo Cuomo



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