

Chapter 6: Chemical Bonds

- When the highest occupied energy level of an atom is filled with e, the atom is stable and not likely to react.
- In other words, all stable elements have 8 electrons in the outer level.
- What determines the chemical properties of an atom?
- How do we know the number of valence electrons an atom has?
- Why do atoms bond?
- Are there different ways atoms can bond?

<u>Chapter 6 Chemical Bonds Wordwise Physical Science</u> <u>Answers</u>

G Psacharopoulos

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Chemical Bonds Phillip Manning,2009 Modern life is made up of a mind boggling array of materials A simple drinking cup for example might be made of Styrofoam paper or glass depending on the drinkers needs at the moment Home storage cabinets can be made of metal wood or plastic Space shuttles are assembled from silicon steel and hundreds of other materials All of these items owe their properties to the chemical bonds between the atoms that make up the substance Chemical Bonds examines the nature of the chemical bonds answering fundamental questions about how they form how they are broken and how they help define life as we know it

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