

Chapter 2. Chemistry Comes Alive (Mastering A&P – Chapter Quiz)

1. When atoms or molecules combine to form larger, more complex molecules, the process is known as _____.
 - a. Catabolism
 - b. Decomposition
 - c. Anabolism
 - d. Exchange
2. A chemical reaction in which bonds are broken is _____.
 - a. Catabolic and endergonic
 - b. Anabolic and endergonic
 - c. Catabolic and exergonic
 - d. Anabolic and exergonic
3. Which of the following factors would speed up the rate of a chemical reaction?
 - a. Lower temperature
 - b. Removal of the biological catalysts
 - c. A high concentration of reagents
 - d. Large particles
4. Which of the following is considered the universal solvent?
 - a. Plasma
 - b. Blood
 - c. Intracellular fluid
 - d. Water
5. In hydrolysis, _____.
 - a. A molecule of water is removed when proteins are synthesized from smaller molecules
 - b. Electrons are lost from a reactant
 - c. A molecule of water is added for each bond broken
 - d. A reactant accepts electrons
6. All electrolytes are _____.
 - a. Polar molecules
 - b. Acids
 - c. Non-polar molecules
 - d. Ions
7. pH is a concentration unit used to measure the concentration of _____.
 - a. Hydroxyl ions in a solution
 - b. Anions in a solution
 - c. Isotopes in a solution
 - d. Hydrogen ions in a solution

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Michael Hoelscher



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