

Section 6.2 Covalent Bonding

(pages 165–168)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy (page 165)

Relating Text and Models As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds (pages 165–167)

- Describe a covalent bond.
- Circle the letters of molecular models that show overlap of atoms overlapping when a covalent bond forms.
 - electron dot
 - structural formula
 - space-filling
 - electron cloud
- Describe a molecule.
- Is the following sentence true or false?
 “In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus.”
- Circle the correct answer. Nitrogen has two valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - zero
 - one
 - two
 - three

Chapter 6 Section 6 2 Covalent Bonding Answer Key

Ralph Thomas Myers



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Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Applied Chemistry for Polytechnic and Engineering Courses** Dr. R.S. Chauhan, 2021-05-27 The book includes the following chapters in details Language of Chemistry Atomic Structure The Periodic table and Atomic properties Water Chemical Bonding Solutions Electrolysis Environmental Chemistry Experiments *Holt Chemistry* Ralph Thomas Myers, 2004 Principles Of Descriptive Inorganic Chemistry Gary Wulfsberg, 1991-05-29 This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts Translated into Italian **Principles of Cell Biology** George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the underlying principles that illustrate both how cells function as well as how we study them It identifies 10 specific principles of cell biology and devotes a separate chapter to illustrate each The result is a shift away from the traditional focus on technical details and towards a more integrative view of cellular activity that is flexible and can be tailored to suit students with a broad range of backgrounds Biological Science Jon Scott, Anne Goodenough, Gus Cameron, Dawn Hawkins, Martin Luck, Jenny Koenig, Alison Snape, Despo Papachristodoulou, Kay Yeoman, Mark Goodwin, 2022 A fresh approach to biology centred on a clear narrative active learning and confidence with quantitative concepts and scientific enquiry Spanning the breadth of biological science and designed for flexible learning it will give you a deeper understanding of the key concepts and an appreciation of biology as a dynamic experimental science **Chemical Interactions**, 2005 Quantitative Chemical Analysis Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a

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