

Revision practice worksheets

Name: _____

Date: _____

1. In which substance is the oxidation number of nitrogen zero?
A. NH_3 B. N_2 C. NO_2 D. N_2O
2. What is the oxidation number of carbon in NaHCO_3 ?
A. +6 B. +2 C. -4 D. +4
3. In the reaction $\text{Al}^0 + \text{Cr}^{3+} \rightarrow \text{Al}^{3+} + \text{Cr}^0$, the reducing agent is
A. Al^0 B. Cr^{3+} C. Al^{3+} D. Cr^0

4. In the reaction $2\text{K} + \text{Cl}_2 \rightarrow 2\text{KCl}$, the species oxidized is
- A. Cl_2 B. Cl^- C. K D. K^+
5. As an S^{2-} ion is oxidized to an S^0 atom, the number of protons in its nucleus
- A. decreases B. increases
C. remains the same

6. Given the probable reaction for the nickel-cadmium battery:
- $$2\text{Ni(OH)}_2 + \text{Cd} \rightleftharpoons 2\text{Ni(OH)}_3 + \text{Cd(OH)}_2$$
- Which species is oxidized during the discharge of the battery?
- A. Ni^{2+} B. Ni^{3+} C.

7. If element X forms the oxides XO and X_2O_3 , the oxidation numbers of element X are
- A. +1 and +2 B. +2 and +3
C. +1 and +3 D. +2 and +4
8. Oxygen has a positive oxidation number in the compound
- A. H_2O B. H_2O_2 C. OF_2 D. IO_2
9. What is the oxidation number of sulfur in H_2SO_4 ?
- A. 0 B. -2 C. +6 D. +4

10. In the equation $\text{Cu(s)} + 2\text{Ag}^+(\text{aq}) \rightarrow \text{Cu}^{2+}(\text{aq}) + 2\text{Ag(s)}$, the oxidizing agent is
- A. Cu^0 B. Ag^+ C. Cu^{2+} D. Ag^0
11. In the reaction $\text{Al} + \text{Cr}^{3+} \rightarrow \text{Al}^{3+} + \text{Cr}$, the reducing agent is
- A. Al B. Cr^{3+} C. Al^{3+} D. Cr
12. In the compound Na_2HPO_4 , which element has a negative oxidation number?
- A. H B. O C. P D. Na

Chapter 2 worksheet Redox Answers

M Mark



Chapter 2 worksheet Redox Answers:

Chapter 2worksheet Redox Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Chapter 2worksheet Redox Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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