



# Chapter 5 - Electrons in Atoms - Outline

## 5.1 - Light and Quantized Energy

### The Atom and Unanswered Questions

- Rutherford proposed that all of an atom's positive charge and virtually all of its mass are concentrated in a nucleus that is surrounded by fast-moving electrons
- Scientists observed that certain elements emitted visible light when heated in a flame
- analysis of the emitted light revealed that an element's chemical behavior is related to the arrangement of the electrons in its atoms

### The Wave Nature of Light

Visible light is a type of **electromagnetic radiation**, a form of energy that exhibits wave-like behavior as it travels through space

### Characteristics of Waves

# Chapter 5 Electrons In Atoms

**Barry G. Hinwood**



## Chapter 5 Electrons In Atoms:

**The Progress of Atomic Theory** Albert Cushing Crehore, 1926      **Basic Chemical Principles** Edward I. Peters, 1988  
An Epitome of Swedenborg's Science Frank Washington Very, 1927      **Introduction to Chemical Principles** Edward I. Peters, 1986      **Monograph Series** American Chemical Society, 1922      *Electricity* Andrew William Kramer, 1929  
*Introduction to the Physics of Highly Charged Ions* Heinrich F. Beyer, Viatcheslav P. Shevelko, 2016-04-19 Emphasizing a physical understanding with many illustrations Introduction to the Physics of Highly Charged Ions covers the major areas of x ray radiation and elementary atomic processes occurring with highly charged ions in hot laboratory and astrophysical plasmas Topics include light and ion sources spectroscopy atomic structure magnetic and      *The Elements of qualitative chemical analysis v. 1* Julius Stieglitz, 1911      *Atomic Collision Theory* B. H. Bransden, 1983      *The Electronic Theory of Valency* Nevil Vincent Sidgwick, 1927      The Structure of Matter John Arnold Cranston, 1924      **Introduction to Modern Physics** Floyd Karker Richtmyer, 1928      **The Structure of the Atom** Stephen Miall, 1922      *A Textbook of Science for the Health Professions* Barry G. Hinwood, 1997 To keep abreast with current developments in medicine members of the health care team require a firm grasp of science to cope with changes in technology and understanding of the mechanisms of body function This is in addition to developing a range of interpersonal and communication skills There are sections covering biology chemistry physics nutrition biochemistry medical microbiology and physiology Highly illustrated it includes over a hundred applications and examples to assist the reader in relating science to health care Throughout the text is divided into units containing a common theme and each chapter contains a list of objectives and a summary      **Intermediate Electricity and Magnetism** Robert Alexander Houstoun, 1928      The Nature of Matter and Electricity Daniel Frost Comstock, Leonard Thompson Troland, 1917      **The Iron Age**, 1925      **Modern Electrical Theory** Norman Robert Campbell, 1907      The Thermionic Vacuum Tube and Its Applications Hendrik Johannes Van der Bijl, 1920      *The Elements of Qualitative Chemical Analysis* Julius Stieglitz, 1911

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