

# Chapter 3 Study Guide

## SECTION 3.1 Matter

In your textbook, read about elements and atomic structure.

Use each of the terms below just once to complete the passage.

atom                      electrons                      element                      neutrons                      nucleus                      protons

A(n) (1) \_\_\_\_\_ is a substance that cannot be broken down into simpler substances. A(n) (2) \_\_\_\_\_ is the smallest particle of matter having all that element's characteristics. It is made up of smaller particles. The (3) \_\_\_\_\_ is made up of protons and neutrons. Small particles that have mass and positive electrical charges are (4) \_\_\_\_\_. Particles that have about the same mass as protons, but that are electrically neutral are (5) \_\_\_\_\_. Surrounding the nucleus of an atom are tiny particles called (6) \_\_\_\_\_, which have little mass, but have negative electrical charges that are exactly the same magnitude as the positive charges of protons.

In your textbook, read about atomic structure and isotopes.

Complete each statement.

7. The number of protons in an atom's nucleus is the \_\_\_\_\_.
8. When atoms of the same element have different mass numbers, they are known as \_\_\_\_\_ of that element.
9. The spontaneous process through which unstable nuclei emit radiation is called \_\_\_\_\_.
10. A(n) \_\_\_\_\_ represents the area in an atom where an electron is most likely to be found.
11. An atom that gains or loses an electron and has an electric charge is called a(n) \_\_\_\_\_.
12. The combined number of protons and neutrons is the \_\_\_\_\_.
13. The \_\_\_\_\_ is the average of the mass numbers of the isotopes of an element.

# Chapter 3 Study Guide For Content Mastery Answer Key

**M Woodhall**



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