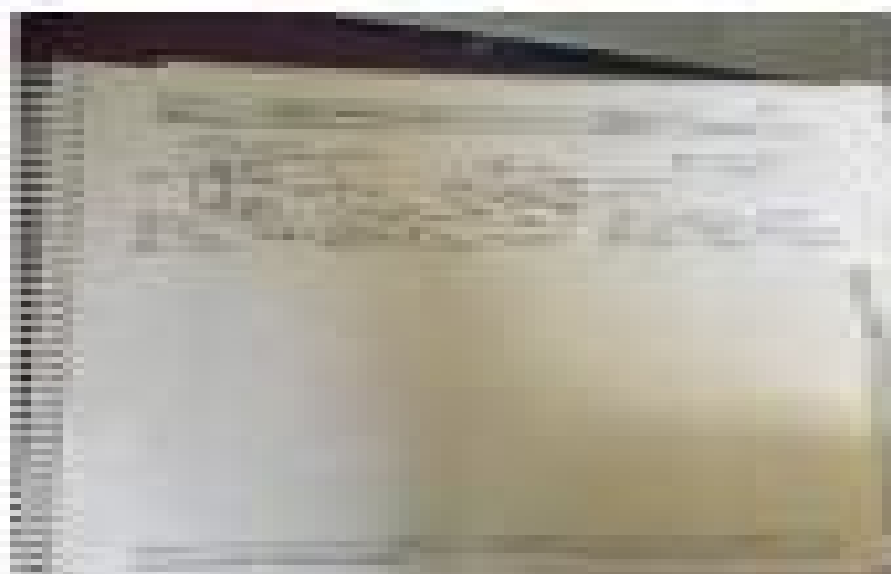


Chemistry 112- Experiment 7: Pre-Lab

Read the Following carefully!

1. When you are selecting what the polyhalides must undergo and then select the base. The rules are pretty straightforward the first element must be a halogen and the second must be a metal. The base must not be a hydroxide, carbonate, or phosphate (very strong bases) and the base must be a metal. The products should be water.
2. When you are asked to write a chemical equation, it should be a simple one. The reaction is that a polyhalide reacts with a base to form a metal halide and water. The products should be water. The products should be water. The products should be water.



3. The products of the reaction should be water and a metal halide. The products should be water and a metal halide. The products should be water and a metal halide. The products should be water and a metal halide.

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