

# CHEM 121 Exam 1 Practice Exam Final Study Guide

## Questions with Reviewed Answers Latest

### 2025/2026 GRADED A+

1. What is the empirical formula of a compound that is 73.14% C, 7.37% H with the remainder oxygen? :  $C_5H_6O$

2. Aluminum reacts with oxygen to produce aluminum oxide.4



If 3.0 moles of Al reacts with excess  $O_2$ , how many moles of  $Al_2O_3$  can be formed? : 1.5 mol

3. The Roman numerals in the reaction given represent the coefficients in the balanced chemical equation.

What are the values of the coefficients?



4. mole: amount

5. kelvin: temperature

6. meter: length

7. A concise verbal statement or a mathematical equation summarizing a broad variety of observations:: law.

8. Which of the following is not the symbol of an element?

A. Cu

B. Ni

C. CO

D. Ag

# Chem 121 Final Exam Answers Ccbc

**Steve DeRoy**



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