

# Qualitative Analysis and Chemical Bonding

## Concepts:

- Chemical bonding
- Covalent bonding
- Ionic bonding
- Metallic bonding
- Intermolecular bonding
- Physical & chemical properties
- Electronegativity

## Introduction

- Study of atoms, molecules, ionic crystals, covalent crystals, polymers, etc.
- In order to understand these all, understanding of bonding is very important. It helps us to explain the physical, chemical, and biological properties of matter.
- In order to study a system, we need to know its structure, energy, and the nature of the forces that hold it together.

### Atomic Structure

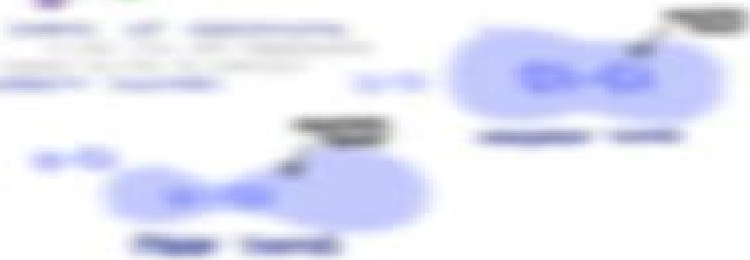
- Atoms are made up of three subatomic particles: electrons, protons, and neutrons.
- Electrons are negatively charged, protons are positively charged, and neutrons are neutral.
- Electrons are present in all atoms, while protons and neutrons are present in the nucleus.
- The mass of an electron is very small compared to the mass of a proton or neutron.



### Chemical Bonding

Chemical bonding is the process by which atoms combine to form molecules or crystals. It involves the sharing or transfer of electrons between atoms.

#### Types of Bonding



Electronegativity is the tendency of an atom to attract a shared pair of electrons towards itself. It is a measure of the ability of an atom to attract electrons.

Electropositivity is the tendency of an atom to lose electrons. It is a measure of the ability of an atom to donate electrons.

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