

ChemQuest 47

The Reaction Quotient

Name: _____

Date: _____

Hour: _____

Critical Thinking Questions

1. If K_c for a given reaction is very large would there be a large amount of products or reactants in the mixture?
Large amounts of products.
2. If K_c for a given reaction is very small would there be a large amount of products or reactants in the mixture?
Large amounts of reactants.
3. Offer a mathematical explanation for your answers to questions 1 and 2.
Since K_c equals the concentration of products divided by reactants, a large K_c means a large numerator (products) whereas a small K_c indicates a small numerator.

Information: The Reaction Quotient

The reaction quotient, Q_c , is calculated in the same way as you would calculate the equilibrium constant. For the reaction $aA + bB \rightleftharpoons cC + dD$, the reaction quotient is:

$$Q = \frac{[C]^c [D]^d}{[A]^a [B]^b}$$

It is important to keep in mind that the reaction quotient does not involve equilibrium concentrations. The concentrations used to calculate Q_c are at any time, not just at equilibrium.

Critical Thinking Questions

4. Consider the following reaction: $CO + 3H_2 \rightleftharpoons CH_4 + H_2O$. While carrying out a reaction between carbon monoxide and hydrogen, a scientist analyzed the mixture and found that in the 3.5 L container there were 0.35 moles of CO, 0.42 moles of H_2 , 0.29 moles of CH_4 , and 0.38 moles of H_2O . What is the reaction quotient for this mixture?

$$[CO] = 0.35 \div 3.5 = 0.10 \text{ M}; [H_2] = 0.42 \div 3.5 = 0.12 \text{ M}; [CH_4] = 0.29 \div 3.5 = 0.0829 \text{ M};$$

$$[H_2O] = 0.38 \div 3.5 = 0.109 \text{ M}$$

$$Q_c = \frac{[CH_4][H_2O]}{[CO][H_2]^3} = \frac{(0.0829)(0.109)}{(0.10)(0.12)^3} = 52.3$$

Information: What Q_c Tells Us

As a reaction proceeds it will always tend to go toward equilibrium. For example, the equilibrium constant for the reaction described in question 4 is 3.92. The concentration of products and reactants will adjust themselves so that as the reaction progresses until the products divided by reactants (raised to the appropriate power) will equal 3.92.

Chem Quest Answer Key

**Corwin Hansch, Peter George
Sammes, John Bodenhan Taylor**

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