

Name: \_\_\_\_\_

There is only one correct answer for each question.

1. Choose the element from the list below.

- (A) sodium chloride      (B) water      (C) helium      (D) rust

2. A slab of lithium weighs 1.49 g and has a volume of 2760.7 mm<sup>3</sup>. What is the density in g/cm<sup>3</sup>? [1000 mg = 1.0 g, 10 mm = 1.0 cm]

- (A)  $5.40 \times 10^{-4}$  g/cm<sup>3</sup>      (B)  $5.40 \times 10^{-4}$  g/cm<sup>3</sup>      (C) 1.65 g/cm<sup>3</sup>      (D) 0.70 g/cm<sup>3</sup>

3. A chemical change

- (A) occurs when methane gas is burned.      (B) occurs when paper is shredded.  
(C) occurs when water is vaporized.      (D) occurs when salt is dissolved in water.

4. All of the following are SI base units of measurement, EXCEPT

- (A) meter      (B) pound      (C) second      (D) kelvin

5. What is the mass of 12.5 moles of SO<sub>2</sub>? [SO<sub>2</sub> molar mass 64.1 g/mol]

- (A) 801 g      (B) 0.195 g      (C) 600 g      (D) 5.13 g

6. How many molecules of PCl<sub>3</sub> are there in 6.8 g of PCl<sub>3</sub>? [PCl<sub>3</sub> molar mass 137.3 g/mol]

- (A)  $6.02 \times 10^{20}$       (B)  $2.0 \times 10^{22}$       (C)  $3.0 \times 10^{22}$       (D)  $6.02 \times 10^{22}$

7. How many significant figures are in 0.00523980 mL?

- (A) 3      (B) 4      (C) 5      (D) 6

# Chem 1035 Placement Test

**Fred Honig**



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