

CHEM 103 – Final Exam Study Guide

Chapters 9-10-11 – Organic Chemistry – 38 questions (76 points)

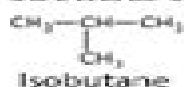
1. Functional groups – all of them

TABLE 10-2 Common Functional Groups

Typical Compound	Functional Group	General Formula	Condensed Formula	Example Structural Formula	Example IUPAC Name	Common Name
Alcohol	Hydroxyl	$R-OH$	$R'OH$	CH_3CH_2-OH	Ethanol	Methyl alcohol
Aldehyde	Carbonyl	$R-C(=O)H$	$R'CHO$	$CH_3-C(=O)H$	Formaldehyde	Formic aldehyde
Ketone	Carbonyl	$R-C(=O)R'$	$R'COR''$	$CH_3-C(=O)-CH_3$	Acetone	Acetone
Carboxylic Acid	Carboxyl	$R-C(=O)OH$	$R'COOH$	$CH_3-C(=O)OH$	Acetic acid	Acetic acid
Ester	Ester	$R-C(=O)OR'$	$R'COOR''$	$CH_3-C(=O)OCH_3$	Methyl acetate	Methyl acetate
Amide	Amide	$R-C(=O)NH_2$	$R'CONH_2$	$CH_3-C(=O)NH_2$	Formamide	Formamide
Nitrile	Nitrile	$R-CN$	$R'CN$	CH_3-CN	Acetonitrile	Nitrile
Alkene	Alkene	$R-CH=CH-R'$	$R'CH=CHR''$	$CH_2=CH_2$	Ethene	Ethylene
Alkyne	Alkyne	$R-C\equiv C-R'$	$R'C\equiv CR''$	$CH\equiv CH$	Ethyne	Ethyne
Alkyne	Alkyne	$R-C\equiv C-R'$	$R'C\equiv CR''$	$CH\equiv CH$	Ethyne	Ethyne
Alkyne	Alkyne	$R-C\equiv C-R'$	$R'C\equiv CR''$	$CH\equiv CH$	Ethyne	Ethyne

2. Recognize the type of compound based on the name-- Above
3. Molecular, Condensed, Expanded, Line formulas – you need to know how to write them all
4. Isomers – structural or constitutional, geometric, conformational

- a. Structural/Constitutional- same formula, different connectivity



- b. Geometric-



- c. Conformational- rotation about carbon-carbon single bonds

5. Saturated / Unsaturated hydrocarbons

- a. Saturated- Only single Bonds
- b. Unsaturated- Has at least one double or triple carbon-carbon bond

6. Aliphatic / Aromatic – recognize

- a. Aromatic- ring structure with alternating double bonds
- b. Aliphatic- everything else

7. Heterocycles – recognize



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Karin Nielsen-Saines



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