

**Unit 3 ~ Learning Guide Name:** Aanya Nilin Shetty
**Instructions:**

Using a pencil, complete the following notes as you work through the related lessons. Show ALL work as is explained in the lessons. You are required to have this package completed BEFORE you write your unit test. Do your best and ask questions if you don't understand anything!

Please print out the Formula Sheet found at the start of the course. You will need this to do assignments and tests!

\* Highly Exothermic reaction are less likely to be in equilibrium as they have a greater  $E_a$  reverse

**Equilibrium:**

1. When does equilibrium occur?

Equilibrium occurs when opposing processes occur at equal rates in a closed system at a constant temperature

\* chemical equilibrium is also known as dynamic equilibrium

2. Give an example of two physical process that oppose each other during equilibrium?

A water solution in a closed container can have a physical equilibrium, and the opposing processes would be evaporation and condensation.

3. Complete the ICE chart for this reaction given that 4.0 moles of  $N_2$  gas and 8.0 moles of  $H_2$  gas are placed into a 1.00 L sealed flask and allowed to come to equilibrium. At equilibrium, 2.4 moles of  $NH_3$  are present.

	$N_{2(g)}$	+	$3H_{2(g)}$	$\rightleftharpoons$	$2NH_{3(g)}$
Initial concentration [I]	4.0 M		8.0 M		0 M
Change in concentration [C]	- 1.2 M		- 3.6 M		+ 2.4 M
Equilibrium concentration [E]	2.8 M		4.4 M		2.4 M

\* Ratio of the balanced equation (1:3:2)

# Chem 12 Study Guide

**Sharon Wynne**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

## Chem 12 Study Guide:

**Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors' approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors' philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors' approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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**Resources in Education**, 1984 *Fundamentals of Chemistry, Study Guide* James E. Brady, John R. Holum, 1988-04-20

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