Intermolecular Forces Worksheet

Answers are on page 3 & 4. Do the problems on your own BEFORE looking at the answers.

1. Predict the molecular shape of each of the following:

a. H₂S

b. CCl₄

e. SO₂

d. BrF

d. PCI₅

 List all types of IMFs that would occur in each of the following (you should have a good enough understanding of electronegativities to answer all of these, except maybe SO₂, without look up the electronegativity numbers).

a. CH₃CF₃

b. CCl₄

c. SO₂

d. BrF

e. (CH₃)₃N

f. PCls

 H₃S, O₂ and CH₃OH all have comparable molecular masses. List the dominant type of IMF for the pure substances, then rank the strength of each compound based on IMFs within the samples.

(1 = strongest, 2 = in between, 3 = weakest).

Substance	IMF	Relative Strength
HBr		
O ₂		
CH ₂ OH		

Chem 150 Activity On Intermolecular Forces Answers

Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe

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