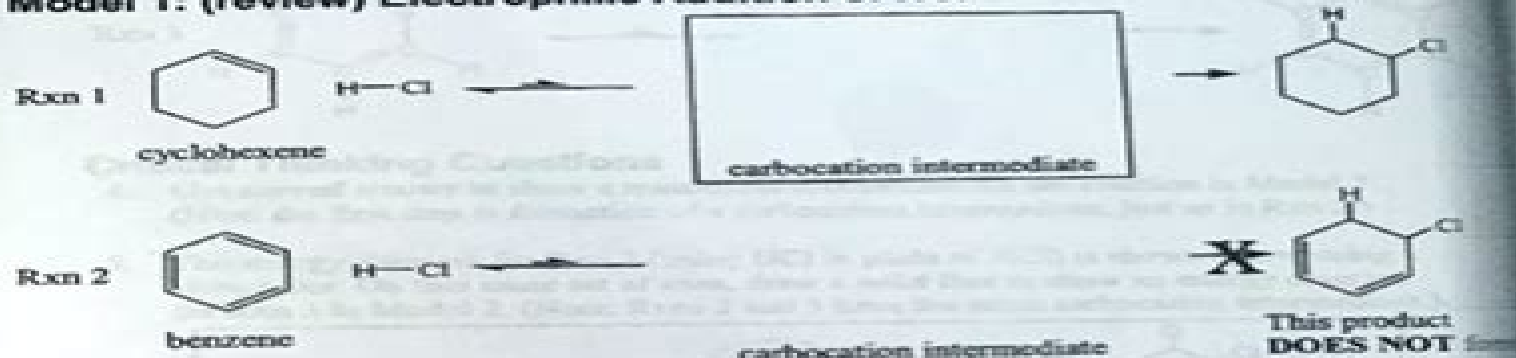


ChemActivity 29

Part A: Electrophilic Aromatic Substitution

(What products are formed when a strong electrophile is added to benzene?)

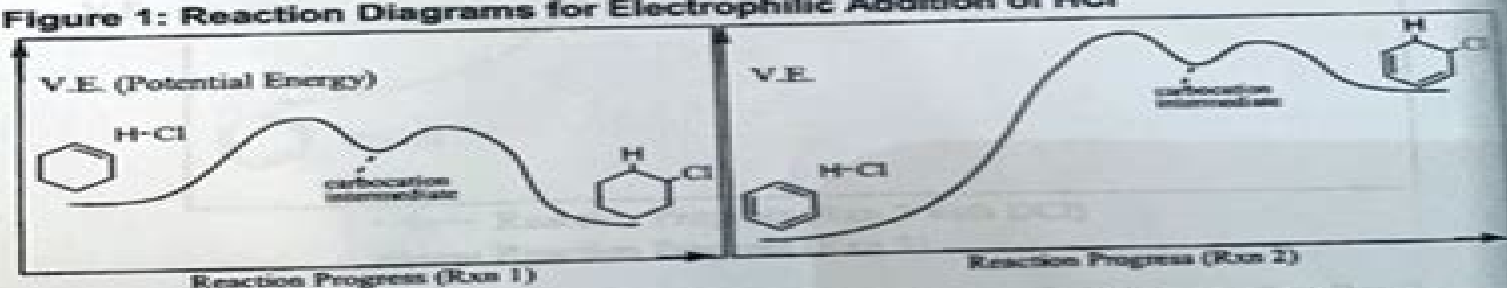
Model 1: (review) Electrophilic Addition of HCl



Critical Thinking Questions

- For Rxn 1 (above) draw curved arrows showing the mechanism of electrophilic addition of HCl. Include an appropriate carbocation intermediate in the box above.

Figure 1: Reaction Diagrams for Electrophilic Addition of HCl



- Rxn 1 is slightly down-hill in terms of energy. Rxn 2 is very up-hill in terms of energy (see Figure 1). Construct an explanation for the large difference in energy between the reactants and the product in Rxn 2.
- Draw the carbocation that would form in Rxn 2. Explain why this carbocation goes back to the starting material (H-Cl and benzene) instead of forming the product.

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[Anonymus AC00033125]



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