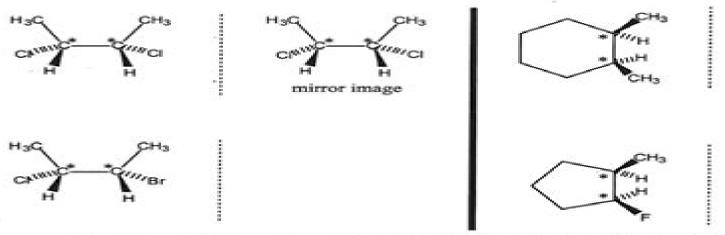
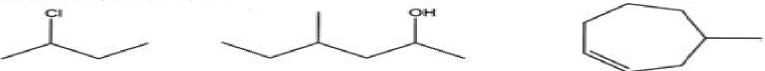
3. Consider the structures below.



- a) Draw the mirror image of each molecule (the first one is done for you.)
- b) Circle the structures that are not chiral and use a dotted line to show any internal symmetry planes. Label these structures meso compound.
- 4. Read the assigned sections in your text and do the assigned problems.

Exercises for Part B

 Sometimes a molecule with one or more chiral centers is drawn without giving enough information to determine if the chiral center/s are R or S. Mark each chiral center below with an *.



By convention, if there is not enough information given to assign R or S, we are to assume that the structure represents an even mixture of all stereochemical possibilities. For each structure above, draw all stereoisomers implied by the drawing. Circle those structures that imply a racemic mixture (racemic mixture = 1:1 mixture of exactly two enantiomers).

Chemactivity 58 Answers

Mike Jess

Chemactivity 58 Answers:

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