CHM 1025- Into to Chem- Spring 2022 EXAM REVIEW PART 1

Rules for Significant Figures

All nonzero numbers are significant.

Zeros:

Zeros between two non-zero digits ARE significant. Ex. 2051(ARE or AREN'T)

Leading zeros AREN'T significant. (ARE or AREN'T)

Trailing zeros to the right of the decimal ARE significant. Ex. 72.000 (ARE or ARENT)

Addition +Subtraction; When adding or subtracting, the answer is limited by the value with the fewest decimal places.

<u>Multiplication + Division</u>: When multiplying or dividing measurements, the answer is limited by the value with the fewest significant digits.

Rounding:

If the first nonsignificant digit < 5, we drop all the nonsignificant digits.

If the first nonsignificant digit is ≥ 5, we increase the last significant digit by 1 and drop all nonsignificant digits.

Add 0.56 g, 0.796 g, and 0.437 g and round off the answer.

1.8 g

2. Multiply 7.605 cm and 2.24 cm and round off the answer.

17.0 cm²

3. If an automobile airbag inflates in 73 µs, what is the time in seconds in scientific notation?

 7.3×10^{-5} s.

4. If 500 mL of liquid mercury weighs 9.85 kg, what is the mass in pounds? (1 lb = 454 g)

21.7 lb

Chemfiesta Final Exam Review

David Baud

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